1	Influence of paste thickness on coated aggregates on properties of
2	high-density sulphoaluminate cement concrete
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11	Abstract: An improved method for the densified mixture design algorithm and Fuller
12	curve were used to design high-density sulphoaluminate cement concrete (HDSC). The
13	performance of HDSC is significantly influenced by the paste thickness on the coated
14	aggregates. Sulphoaluminate cement concrete mixtures containing aggregates coated
15	with 3 different paste thickness of t=10 $\mu$ m, 20 $\mu$ m, and 30 $\mu$ m and water-binder ratios
16	(W/B) of 0.25, 0.30 and 0.35 were prepared. The results of experiments show that
17	paste thickness on the coated aggregates significantly influences the mechanical
18	properties and durability of HDSC. With the increase of paste thickness, the
19	compressive strength is increased, but the electrical resistivity is decreased, particularly
20	at the early ages of 1 and 3 days. The sulfate corrosion resistance coefficients of HDSC
21	are larger than 1.0, the total porosity can be less than 7%, and the micropore (i.e. with
22	pore size less than 20nm) can be larger than 70%.
23	Keywords: sulphoaluminate cement concrete; Fuller curve; densified mixture design
24	algorithm; coating thickness; durability

# 25 **1. Introduction**

Sulphoaluminate cement is a type of "low energy" cement compared to Portland cement [1], possessing advantageous properties such as high early-age compressive strength, short setting time and shrinkage compensation and it is typically used in the

29 marine engineering field [2-4]. However, as a special type of cement, the mechanical properties and durability [5-6] of sulphoaluminate cement concrete (SACC) has not 30 been well studied. Under harsh environments, the harmful external ions and water can 31 easily permeate into the concrete interior, destroying its structure and shorten its 32 service life. However, a compact concrete structure can lead to improved strength and 33 durability. The importance of pore structure and its impact on durability has been 34 highlighted in numerous studies [7]. Many researchers also found that the concrete 35 36 pore structure improved the interfacial transition zone (ITZ) and dominated engineering properties, such as strength and durability [8-9]. For such reasons, 37 high-density concrete has been widely used to achieve outstandingly durable concrete 38 39 structures.

However, it must be noted that little work has been conducted on SACC mixture 40 design as a high-density concrete. Therefore, the major work required is designing an 41 appropriate mix proportion to produce the high-density sulphoaluminate cement 42 concretes (HDSCs). The densified mixture design algorithm (DMDA) is derived from 43 44 the maximum density theory and excess paste theory, proposed by Hwang et al. [10-12]. This method is based on the hypothesis that the physical properties can be 45 optimized when the packing density is high. The major difference from the other 46 mixture design algorithms is that instead of partial replacement of cement, DMDA 47 48 incorporated the use of fly ash to fill the void between aggregates and hence increase the density of the aggregate system. In such a way, the cement paste content can be 49 reduced without affecting the other properties such as workability, and strength [13]. 50 Lots of research [13-16] shows that it is feasible to produce the eco-friendly 51 lightweight concrete, 52 construction bricks, high-performance concrete and 53 self-compacting concrete using the DMDA method with the incorporation of an admixture, such as fly ash or slag powder. However, to simplify the derivation, it is 54 necessary to assume that the aggregate is spherical, which is physically very hard to 55 56 achieve and thus gives rise to errors.

57 In additional, it is commonly thought that the cement paste volume is a key factor 58 in achieving a desirable concrete workability and durability [17-19]. A work studied

59 the effect of cement content on transport processes important to the durability of concrete structures, such as electrical conduction and chloride diffusion. It was found 60 that the resistance to transport reduced as cement content was increased [20]. Hwang et 61 al. proposed a particular DMDA, in which the concept and formula of the paste 62 thickness on coated aggregates were introduced. A complete and precise formula to 63 estimate the optimum coating thickness on the aggregates was derived to ensure the 64 use of sufficient coating paste and a dense concrete structure is obtained [21]. Kolias 65 66 and Georgiou studied the effect of paste volume and of water content on capillary absorption and strength on concrete mixes. It is found that strength increases and 67 capillary absorption decreases when the volume of the water or the volume of the paste 68 decreases [22]. Chen et al. demonstrated that the paste thickness on the coated 69 70 aggregates has a positive effect on the slump flow, concluding that a thickness of  $42\mu m$ 71 produced self-compacting concrete which flowed to a diameter of 680mm [23]. Last but no least, using less cement reduces energy consumption and CO<sub>2</sub> emissions 72 73 associated with its production process.

74 In this paper, an improved DMDA method is developed to simplify the calculation process. Introducing the assumption that the aggregates are square and 75 spherical in shape allows a more accurate engineering design requirement to prepare 76 HDSC. The Fuller curve was used to calculate the aggregate gradation, and the sieve 77 78 analysis was used to calculate the specific surface area of aggregates. Sulphoaluminate cement, replaced by approximately 5% superfine slag powder, and 10% fly ash, both 79 by weight, was used as the cementitious materials for preparing the HDSC and the 80 improved DMDA calculated the dosage of cementitious material. Finally, the effects of 81 82 paste thickness around the aggregates on mechanical and durability properties of 83 HDSC were investigated.

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#### 85 **2. Raw materials and methods**

#### 86 2.1 Sulphoaluminate cement and admixtures

87 Sulphoaluminate cement of strength class 42.5, fly ash and slag powder were 88 imported from mainland China. The chemical compositions of the sulphoaluminate cement, fly ash and slag powder are shown in Table 1. The average particle size of fly ash and slag powder are  $14.34\mu m$  and  $2.98\mu m$ , respectively. The superplasticizer (SP) used was a polycarboxylate polymer, and its water-reducing capacity in SACC was over 20%.

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 Table 1 Chemical composition of raw materials (wt.%)

Materials	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	CaO	MgO	Fe <sub>2</sub> O <sub>3</sub>	Other	Loss
Sulphoaluminate	11.41	27.87	43.86	-	2.59	13.11	1.16
Fly ash	50.55	29.01	6.00	5.44	-	4.12	2.08
Slag powder	29.46	17.44	34.71	11.02	-	0.3	0.30

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# 95 2.2 Aggregates

River sand (0.075-2.36mm in size) was used as fine aggregates with an apparent 96 density of  $2710 kg/m^3$ , and crushed natural stone (2.36-16mm in size) was applied as a 97 coarse aggregate with an apparent density of  $2740 kg/m^3$ . The aggregate mix 98 99 proportions were also key to improving the packing density of concrete [24]. Fuller 100 curve was based on defining conventional concrete dosages by selecting coarse and 101 fine aggregate proportions according to the adjustment within the standard curve that 102 allows for the maximum compaction of granular elements, which is the method that corresponds to the Gessner parabola [25]. Since the Fuller curve was proposed, it has 103 104 been used for designing concrete mixes for many applications, particularly for those of high-density and high-performance concrete [26-28]. The Fuller curve is a series of 105 curves, widely used for the optimization of concrete aggregates, and expressed as: 106

107  $U(j) = 100 \times (j/D_{max})^{h}$  (1)

where U(j) is the total volume percent of particles passing through a sieve, (%);  $D_{max}$  is the maximal size of the aggregate, (*mm*); *j* is the diameter of the particular sieve, (*mm*); and *h* is the exponent of the equation.

111 The value for *h*, which varies from 0.33 to 0.45, was selected as 0.33 [26, 29] in 112 this study. The mass ratios of aggregates of different particle size are given in Table 2, 113 calculated using Equation (1).

114

Grain size (mm)	0.075-0.15	0.15-0.30	0.30-0.60	0.6-1.18	1.18-2.36	2.36-4.75	4.75-9.50	9.50-16.0
wt.%	0.13	0.17	0.21	0.27	0.35	0.47	0.43	0.39

## 116 **2.3 Methods**

### 117 2.3.1 Concrete samples preparation

The concrete samples were made according to Chinese national standard for testing fresh concrete GB/T50080-2002 [30] (equivalent to ASTM C192M-02). All concrete samples, measured  $100mm \times 100mm \times 100mm$ , were cured at  $20\pm2^{\circ}$ C in molds for the first 24h, then demoulded and cured in an environment of  $20\pm2^{\circ}$ C and at 95±5% RH until the day of testing.

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# 124 2.3.2 Compressive strength test

The compressive strength test of concrete was carried out according to the Chinese National Standard for testing mechanical properties of concrete GB/T 50081-2002 [31] (equivalent to ASTM C39). The compressive strength of concrete was evaluated for the ages of 1, 3 and 28 days at a loading rate of 0.5*MPa/s* as per GB/T 50081-2002.

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# 131 2.3.3 Electrical resistance test

In this study, the concrete was mixed with a water-binder ratio (0.25, 0.30 and 0.35) and cast into 100mm<sup>3</sup> cubes for the electrical resistance test. The negative and positive copper electrodes were placed parallel to each other inside the cubic concrete samples, and the average testing results of the three samples were taken as the representative value. The schematic diagram of the concrete specimen prepared for electrical resistivity measurement is shown in Fig.1.



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Fig.1 Schematic diagram for concrete electrical resistivity measurement

# 141 2.3.4 Sulfate attack resistance

142 The resistance of concrete to sulfate attack was conducted according to Chinese national standard for testing durability of concrete GB/T 50082-2009 [32] (equivalent 143 to ASTM C 1012). The solution was made by dissolving reagent grade sodium sulfate 144  $(Na_2SO_4)$  in deionized water and contained a final  $SO_4^{2-}$  concentration of 33,800ppm 145 (i.e. 5% Na<sub>2</sub>SO<sub>4</sub>). All specimens were stored in plastic containers having the solution 146 with ample space between them. The containers with the specimens were stored in a 147 148 constant temperature  $(20\pm1^{\circ}C)$  room and the solutions were replenished periodically 149 once a week to remain the designated concentration. Other control concrete cubes were kept in deionized water as well. The degree of sulfate attack was evaluated by 150 151 measuring the compressive strength of concrete samples at 28 days, and the ratio of compressive strength was calculated by Equation (2) as follows: 152

$$K_f = \frac{f_{cn}}{f_{c0}} \tag{2}$$

where,  $K_f$  is the compressive strength ratio, (%);  $f_{c0}$  is the average compressive strength (in *MPa*) of the control concrete cubes cured for 28 days in deionized water; and  $f_{cn}$  is the average compressive strength (in *MPa*) of three concrete cubes immersed in 5% sodium sulfate solution for 28 days.

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#### 159 **3. Dosage of cementitious materials**

## 160 **3.1 Packing model of concrete**

According to classical concrete mixture proportion design, the aggregates are considered the main skeleton of concrete and the paste requirements for workable

concrete are determined by the aggregate gradation. Effective packing can be attained 163 164 by selecting accurate proportions of small size particles to fill in the voids between the bigger particles. The Fuller model is considered to be an effective method for 165 preparing high-density concrete by adjusting the particle size for optimum aggregate 166 167 packing [33-34]. The concrete model structure is shown in Fig.2. In the diagram, the 168 aggregates, shown in black, are representing the concrete skeleton, the paste coating 169 them is shown in gray, and the white area among the aggregates are the voids.



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Fig.2 Diagrammatic model of aggregates and paste in concrete.

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#### 3.2 Calculation process of cement paste dosage 173

174 (1) The calculation process of the specific surface area of aggregates

175 The aggregate specific surface area calculation process was simplified by assuming all aggregates to be either spherical or square. The volume of each particle is 176 177 shown in the Equation (3):

178 Spherical:  

$$\frac{m}{\rho} = V = \frac{4}{3} \pi \left(\frac{D}{2}\right)^3 = \frac{1}{6} D \cdot \pi D^2 = \frac{1}{6} D \cdot A$$
(3)  
179 Square:  

$$\frac{m}{\rho} = V = D^3 = \frac{1}{6} D \cdot 6D^2 = \frac{1}{6} D \cdot A$$

Then, according to the definition of the specific surface area with the combination 180 of Equation (3), Equation (4) could be derived as following. 181

182 
$$\frac{n \cdot A}{n \cdot m} = \frac{A}{D \cdot A \cdot \rho} \cdot 6 = \frac{6}{D \cdot \rho}$$
(4)

183 where *m* is the mass of each aggregate particle, (in *g*);  $\rho$  is the apparent density of 184 aggregates, (in  $kg/m^3$ ); *V* is the volume of each aggregate particle, (in  $cm^3$ ); *D* is the 185 particle size of aggregate, (in *mm*); *A* is the specific surface area of each aggregate 186 particle, (in  $cm^2$ ); and *n* is the number of aggregate particles.

187 The aggregate is of varying particle size, so the specific surface area of the 188 aggregates could be achieved according to Equation (5):

189 
$$\mathbf{A}_{s} = \frac{\sum_{i} \left(\frac{6000}{D_{i} \cdot \rho} m_{i}\right)}{M} = \frac{6000}{\rho} \cdot \sum_{i} \left(\frac{K_{i}}{D_{i}}\right)$$
(5)

where  $A_s$  is the specific surface area of aggregates, (in  $m^2/kg$ );  $D_i$  is the intermediate particle size of each aggregate gradation, (in *mm*); *M* is the total mass of aggregates, (in *kg*); *m<sub>i</sub>* is the total mass of each aggregate gradation, (in *kg*); and *K<sub>i</sub>* is the mass fraction of each aggregate gradations, i.e.  $m_i/M$  in %. Based on calculation, the specific surface area of aggregates with different gradations is shown in Table 3.

199

Table 3 Specific surface area of aggregates

Gradation ( <i>mm</i> )	$D_i(mm)$	<i>K</i> <sub>i</sub> (%)	$A_s (m^2/kg)$
0.075-0.15	0.11	4.5	
0.15-0.30	0.23	6.0	
0.30-0.60	0.45	7.8	5.73
0.60-1.18	0.92	10.0	
1.18-2.36	1.77	12.7	
2.36-4.75	1.20	17.1	
4.75-9.50	7.13	21.7	0.71
9.50-16.0	12.75	20.4	

196 (2) The calculation process of the total paste volume

197 Supposing that the average paste thickness of coated aggregates was termed as t198 (in  $\mu m$ ), the dosage of paste was calculated using Equations (6) and (7):

$$V = \mathbf{t} \cdot (M_s A_s + M_G A_G) + V_h \tag{6}$$

where *V* is the unit volume of the paste, (in  $m^3$ ); *t* is the average paste thickness on the coated aggregates, (in  $\mu m$ );  $M_S$  is the dosage of fine aggregates, (in kg);  $M_G$  is the dosage of coarse aggregates, (in kg);  $A_S$  is the specific surface area of fine aggregates, (in  $m^2/kg$ );  $A_G$  is the specific surface area of coarse aggregates, (in  $m^2/kg$ ); and  $V_b$  is the volume of paste required to fill the pores between the aggregates, (in  $m^3$ );

205 
$$V_{b} = \frac{M_{s}}{\rho_{s}} \cdot P_{s} = \frac{M_{s}}{\rho_{s}} \cdot (1 - \frac{\rho_{s}}{\rho_{s}'})$$
(7)

where  $\rho_s$  is the packing density of fine aggregates, (in  $kg/m^3$ );  $\rho_s$  is the apparent density of fine aggregates, (in  $kg/m^3$ ); and  $P_s$  is the porosity of fine aggregates, (in %).

Based on the concept of particle packing, well-graded aggregates have fewer 208 voids among particles than poorly graded aggregates requiring less cement paste to fill 209 210 the voids. Thus, the additional amount of cement paste remaining can be used to coat the aggregate particles and improve the concrete fluidity. For the same reason, an 211 increase in the volume fraction of aggregates generally results in a reduced concrete 212 fluidity. A high-density concrete with desirable workability is obtained when a suitable 213 amount of cement paste is provided to fill the spaces among the aggregates. The 214 215 dosages of cementitious materials with coating thickness of t=10 $\mu$ m, 20 $\mu$ m and 30 $\mu$ m [35-36] are shown in Table 4, respectively. The water-binder ratio was fixed at 0.25, 216 0.30 and 0.35 after a series of trial mixes, satisfying the requirements of concrete with 217 218 a slump of 250±10mm, respectively.

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Table 4 Water-binder ratio and dosages of cementitious material in HDSCs

N.	Water-Binder	Coating	Cement	FA	SS	SP
INO.	ratio	thickness	$(kg/m^3)$	$(kg/m^3)$	$(kg/m^3)$	$(kg/m^3)$
A1		t=10µm	324	74	39	4.37
B1	0.25	t=20µm	382	88	47	5.17
C1		t=30µm	441	101	54	5.96
A2		t=10µm	299	69	36	4.04
B2	0.30	t=20µm	353	81	43	4.77
C2		t=30µm	407	94	50	5.51
A3		t=10µm	278	64	34	3.76
B3	0.35	t=20µm	328	75	40	4.43
C3		t=30µm	378	87	46	5.11

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# 4. Results and Discussions on Properties of HDSC

222 4.1 Compressive strength



224 Fig.3 Relationship between pastes thickness on the coated aggregates and compressive strength of HDSCs 225 Fig.3 shows the compressive strength of HDSC at different water-binder ratios. It can be found that for all three different water-binder ratios, the compressive strength of 226 concrete increased with the increase of paste thickness on the coated aggregates, 227 228 especially at the early age of 1 and 3 days. Compared with A1, the compressive 229 strength of B1 increased 23%, 8% and 6%, and C1 increased 37%, 17% and 12% at the curing ages of 1, 3 and 28 days respectively. The reason is that the sulphoaluminate 230 231 cement typically attains high early age strength, which is a result of the cement hydration and increases the interfacial bonds in concrete. Table 4 shows that the 232 cement dosage increased for the same coating thickness. 233

At a water-binder ratio of 0.30 and 0.35, concrete samples at 28 days attained very 234 simmilar compressive strengths for aggregates with a coating thickness of 20µm and 235 236 30µm. Although the general trend shows the concrete compressive strength increasing with increase of the paste thickness on the coated aggregates, the compressive strength 237 of B2 is close to that of C2 and the compressive strength of B3 is higher than that of 238 239 C3. The reason can be mainly attributed to the paste thickness on the coated aggregates which improved the structure of the interface transition zone (ITZ) between cement 240 paste and the aggregates. With the decrease of the coating thickness, the space of 241 crystal growth decreases, and the gel could improve the cohesive strength and degree 242 of density [37-38]. The SEM images of concrete A3, B3 and C3 are shown in Fig.4, 243 244 and it can be seen clearly that the hardened cement paste has plenty of gel (C-S-H gel and Al(OH)<sub>3</sub> gel ), fine AFt and little Ca(OH)<sub>2</sub>. The hydration of sulphoaluminate 245

- 246 cement can produce plenty of gel and AFt in B3, leading to a more compacted ITZ
- compared with A3 and C3.



Fig.4 SEM images of interfacial transition zone (ITZ) for HDSCs at 28 days

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## 251 4.2 Electrical resistivity

Several studies [39-40] consider electrical resistivity of concrete is an important factor that indicates the permeability of concrete to aggressive agents such as chloride and carbon dioxide, because electrical resistivity has a strong correlation with the concrete microstructure. The electrical resistivity was decided by the pore solution, which provided a path of conductive ions. This improved conductive performance, and accelerated the corrosion of concrete cubes [41-42].

The electrical resistivity results for HDSCs are shown in Fig.5. The electrical 258 resistivity of all the samples increased with the rise of the curing age, particularly at the 259 260 ages of 1 and 3 days. This can simply be explained by the fact that sulphoaluminate cement has the characteristic of rapid hydration and hardening, and the porosity of 261 hardened cement paste is higher than that of the aggregates. With the increase of paste 262 coating thickness on the coated aggregates, the electrical resistivity of HDSCs 263 decreased at all three water-binder ratios. The electrical resistivity of A1 was 44%, 264 33% and 13% higher than that of C1 at the ages of 1, 3 and 28 days, respectively. 265 Sample A2 was 42%, 35% and 23% higher than that of the sample C2 at the ages of 1d, 266 3 and 28 days, respectively. From Table 4, the dosage of cement increased with the 267 increase of the paste thickness on the coated aggregates. So porosity of concrete 268 269 relatively increased with the increase of the sulphoaluminate cement dosage at the same water-binder ratio. 270



Fig.5 Effect of the paste thickness on the coated aggregates on electrical resistivity of HDSC

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# 274 **4.3 Sulfate attack resistance**

275 The coefficient of sulfate attack resistance is a key parameter in measuring the durability of concrete hence the resistance coefficients of HDSCs were calculated 276 using Equation (2). The coefficients of sulfate attack resistance at the curing age of 28 277 days are shown in Fig.6. The sulfate attack resistance coefficients of HDSCs were 278 279 higher than 1.0, which means that the compressive strength of concrete samples cured in the 5% Na<sub>2</sub>SO<sub>4</sub> solutions are higher than of those cured in the water at 28 days, 280 indicating that former samples have much better capability of resisting sulfate attack. 281 With the increase of paste thickness on the coated aggregates, the sulfate attack 282 283 resistance coefficients decreased evidently at the same water-binder ratio. The coefficient of A1 was 10% higher than that of C1, and A3 was 6% higher than that of 284 C3. Sulfate attack resistance is a very important property of concrete and many studies 285 286 have found that the use of an excessive amount of cement, or too much water, can 287 result in high permeability which has a negative effect on the durability [14, 15, 43]. When the amount of paste thickness on the coated aggregates was appropriate, the 288 289 structure of concrete could reach up to the close packing. Using too little paste was not 290 enough to fill the voids between the aggregates and too much paste would break the 291 close packing structure, which can be used to explain the reduced sulfate attack 292 resistance of C3. Therefore, the proper dosage of cement can also be used to produce



Fig.6 Sulfate attack resistance of HDSCs at a curing age of 28 days

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#### 4.4 Pore structure of HDSCs 297

298 The pore structure of A1, B1 and C1 at a curing age of 28 days is shown in Figure 7, Figure 8 and Table 5. Figures 7 and 8 showed that the number of micropore (less 299 than 20nm) and macropore (more than 200nm) was more than 70% and less than 10% 300 respectively. The cumulative volume of pores in A1 was much more than that in C1, 301 302 and the volume of micropore of the former was lower than that of the latter. These pore structure results are in accordance with the results of electrical resistivity  $(t=10\mu m)$  in 303 304 Fig.5 and sulfate corrosion resistance ( $t=10\mu m$ ) in Fig.6.









Table 5 Pore structure characteristic parameters of HDSCs

	Damasitas (0/)	Pore Volume	Specific Area of	Average Pore	
	Porosity (%)	(ml/g)	Pores $(m^2/g)$	Size ( <i>nm</i> )	
A1	5.0831	0.0210	6.783	12.4	
B1	6.1891	0.0259	8.457	12.2	
C1	6.9700	0.0296	10.014	11.8	

# 312 **5. Conclusion**

(1) HDSCs were designed using an improved DMDA method and Fuller curve.
Investigations based on compressive strength, electrical resistivity, sulfate attack and
pore structure analysis indicate that the paste thickness on the coated aggregate is a key
parameter affecting properties of sulphoaluminate cement concrete.

317 (2) The compressive strength of HDSCs increased with the increase of paste 318 thickness on the coated aggregates, particularly at the early stages of curing, namely 1 319 and 3 days. When paste thickness on the coated aggregate was  $10\mu m$ , the compressive 320 strength increased 23% and 8% compared with  $20\mu m$ , and increased 37% and 17% 321 compared with 30µm at the curing ages of 1 and 3 days, respectively.

322 (3) With the increase of paste thickness on the coated aggregates, the electrical 323 resistivity of HDSCs decreased at the same water-binder ratio. When the paste 324 thickness on the coated aggregate was  $10\mu m$ , the electrical resistivity of HDSC was 325 44%, 33% and 13% higher than that of sample with  $30\mu m$  at the ages of 1, 3 and 28 326 days, respectively.

327 (4) The sulfate attack resistance coefficients of HDSCs at 28 days were higher 328 than 1.0, suggesting that concrete samples cured in the 5% Na<sub>2</sub>SO<sub>4</sub> solutions have 329 much better capability of sulfate attack resistance. With the increase of paste thickness 330 on the coated aggregates, the sulfate attack resistance coefficients decreased evidently. 331 At the same water-binder ratio, the coefficient of HDSCs with  $10\mu m$  was 10% higher 332 than that of HDSCs with  $30\mu m$ .

(5) At 28 days, the porosity, pore volume and specific pore area of HDSCs showed
an increasing trend with an increasing paste thickness on the coated aggregates. The
porosity of HDSCs was below 7% with 70% micropore (<20*nm*), and the average pore
size was below 12.4*nm*.

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