

# Development of low drying shrinkage foamed concrete and hygro-mechanical finite element model for prefabricated building façade applications

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## Abstract

Prefabricated lightweight concrete building façade can improve the energy efficiency of buildings and reduce the carbon emission of transportation. However, it is essential to maintain the dimensional stability of the full scale element. The drying shrinkage of lightweight foamed concrete was investigated in this study. The hypothesis of using the drying shrinkage of normal weight concrete to approximate that of lightweight foamed concrete of dry density about 1,500 kg/m<sup>3</sup> counterpart was verified. Three different strategies of reducing drying shrinkage were studied. The drying shrinkage of common ingredients of ordinary Portland cement (OPC) and ground granulated blast-furnace slag (GGBS) was commonly up to 2,000-3,000  $\mu\epsilon$ . The use of magnesium expansive agent with different calcination conditions could not reduce the drying shrinkage. The use of calcium sulfoaluminate

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(CSA) cement with OPC and GGBS could significantly reduce the drying shrinkage within  $1,000 \mu\epsilon$  in standard testing environment. The formulation developed in laboratory was scaled up in a concrete production plant for prefabricated concrete elements. A lightweight full scale panel (the wet density was about  $1,700 \text{ kg/m}^3$ ) was fabricated. The drying shrinkage of the developed formulation with CSA cement was only  $161 \mu\epsilon$  in the field test. A hygro-mechanical model was developed to model the diffusion, shrinkage and plastic strain evolution. The incremental stress-strain constitutive relationship of the hygro-mechanical model was derived for incorporating it into general finite element routine. The model parameters were calibrated by the drying shrinkage measurements in this study. The calibrated model demonstrated the cracking potential of three typical reinforced concrete panels of three different formulations studied in this study.

*Keywords:*

foamed concrete; drying shrinkage; calcium sulfoaluminate cement; magnesium expansion agent; hygro-mechanical model; finite element

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## 1. Introduction

Residential buildings in densely populated regions are commonly made of reinforced concrete. About 40-50% of energy consumed in buildings is spent on space heating and cooling. Heat is wasted through the building envelope. Even inside the building, heat may be transferred from one compartment to another unintentionally. The problem is exacerbated when floor heating system is used [57]. To improve the energy efficiency of building, it is desirable to minimise the heat transfer through the building envelope and partition of

9 compartments by reducing the thermal conductivity, which is defined by the  
10 product of thermal diffusivity, specific heat capacity and density. As a general  
11 rule, the lower the density is, the lower the thermal conductivity for the  
12 same type of material. The density of concrete or cementitious material can  
13 be reduced by using lightweight aggregate [3, 35], incorporating significant  
14 volume of air void (aerated concrete) [28, 59] or the combination of both  
15 [54, 51, 39, 47]. The air void of aerated concrete can be incorporated by  
16 gas-forming chemicals (aluminium powder, hydrogen peroxide, potassium  
17 permanganate or calcium carbide) or preformed foam by mixing compressed  
18 air, pressurised water and foaming agent (detergents, resin soap, saponin or  
19 hydrolysed proteins) [52, 53, 43]. Alternatively, if the targeted dry density is  
20 higher than  $1,200 \text{ kg/m}^3$ , it is possible to mix foaming agent in the wet mix  
21 to incorporate sufficient air void in the matrix by the shear stress induced  
22 during mixing. However, the dosage of foaming agent to achieve the targeted  
23 density depends on the type of foaming agent, rheology of the mix, mixer  
24 type and mixing time [17].

25 There are satisfactory solutions of internal non-structural partition walls  
26 such as autoclaved aerated concrete blocks and lightweight gypsum blocks.  
27 However, they are not suitable for the external walls and floor slabs which are  
28 usually structural elements. An alternative is to reduce the density of normal  
29 concrete of the building envelope and floor slab. Nevertheless, it is more  
30 difficult to control the quality of cast-in-situ lightweight concrete because it is  
31 sensitive to temperature, member geometry and casting procedure. Instead,  
32 the quality assurance can be improved by prefabrication in factory. Moreover,  
33 it is more environmental friendly and the productivity is higher to adopt

34 prefabricated reinforced concrete elements compared to traditional cast-in-  
35 situ method [16, 31, 14].

36 The authors used OpenLCA 1.6.3 with the European reference Life Cy-  
37 cle Database (ELCD) to estimate the Global Warming Potential (GWP)  
38 of the transportation of twelve  $2.9\text{ m}\times 2\text{ m}\times 0.06\text{ m}$  prefabricated reinforced  
39 concrete walls with different specific gravity for 200 km from the factory to  
40 construction site by a typical lorry. The GWP is reduced by 44% when the  
41 density of the material is two-third of normal reinforced concrete while it is  
42 75% less when the density is reduced by half (the density is assumed to be  
43  $2,400\text{ kg/m}^3$ ) (Figure 1).

44 While it is more effective from energy efficiency point of view to adopt  
45 lightweight concrete for building envelope and floor slab, the structural en-  
46 gineers may concern the long term structural performance and durability. A  
47 compromise is to use lightweight concrete for prefabricated non-structural  
48 permanent formwork of the building envelope and shallow deck of floor slab  
49 while the structural wall or floor slab can be cast-in-situ with normal rein-  
50 forced concrete (Figure 2). The lightweight permanent formwork can reduce  
51 the thermal conductivity of the external wall or slab significantly. Suppose  
52 the thickness and thermal conductivity of the lightweight permanent form-  
53 work ( $1,500\text{ kg/m}^3$ ) and the reinforced concrete wall is 60 mm,  $0.5\text{ W/m}\cdot\text{K}$ ,  
54 180 mm and  $1.3\text{ W/m}\cdot\text{K}$ , respectively. The U-value of a normal reinforced  
55 concrete/lightweight permanent formwork composite is about 46% and 28%  
56 lower than normal concrete wall of thickness of 180 mm and 240 mm, respec-  
57 tively.

58 When the moisture gradient in concrete is positive towards the environ-

59 ment, the evaporable (non-chemically bonded) pore water in the specimen  
60 will diffuse to the surfaces and evaporate. This drying process results in mois-  
61 ture loss and shrinkage. Depending on the level of pore relative humidity ( $h$ ),  
62 one or a combination of the following drying-shrinkage mechanisms: capillary  
63 pressure, disjoining pressure, surface tension, pore blocking, and movement  
64 of interlayer water, can be activated. In the medium to high range of  $h$  (50%-  
65 85%), the shrinkage is attributed to the coaction of the changes in capillary  
66 and disjoining pressures during the drying process [22, 26, 5]. A concave-  
67 curved meniscus is formed in the pores due to moisture loss. The resulted  
68 change in the capillary pressure will compress the solid skeleton and lead  
69 to volumetric contraction. The moisture loss can also reduce the disjoining  
70 pressure in the areas of hindered water adsorption, which in turn decreases  
71 the separation between the solid surfaces. When  $h$  is above 85%, the move-  
72 ment of the evaporable water in the gel pores can be effectively blocked or  
73 slowed down by the link-bottle effect [26] and that is the major cause of  
74 the hysteresis of sorption isotherms [46]. In the low range of  $h$  (<50%), the  
75 meniscus formation in the pores is unstable and the associated capillary pres-  
76 sure effect would become inactive. When drying occurs in this low range, the  
77 decrease of the disjoining pressure and increase of the surface tension between  
78 the cement gel particles [22, 56] are the major mechanisms for the shrinkage.  
79 When drying occurs below 25% , the interlayer water adsorbed between CSH  
80 sheets can be removed and a more compact (i.e. reduction in volume) pore  
81 structure is formed [25, 26].

82 There are extensive review on the mechanical properties and thermal con-  
83 ductivity of lightweight aggregates and foamed concrete, which is referred to

84 cementitious mortar without coarse aggregates and it is the adopted termi-  
85 nology in this study, however, little investigation on the drying shrinkage  
86 based on different formulation of the mix is available in literature [43]. The  
87 reported drying shrinkage of lightweight aggregates concrete ranges between  
88 600 and 1,200  $\mu\epsilon$  that depends on the aggregate type, aggregate content and  
89 initial saturation of the aggregates [1, 29, 20]. The drying shrinkage of foamed  
90 concrete ranges from 600  $\mu\epsilon$  to 3,000  $\mu\epsilon$  [43].

91 In this paper, different approaches of reducing the drying shrinkage of  
92 foamed concrete was investigated. The same approach is applicable for  
93 lightweight aggregates concrete. The formulation of foamed concrete was  
94 scaled up with the partnership of a concrete producer in a prefabricated  
95 yard and the drying shrinkage was compared with the existing strategies of  
96 the concrete producer. In addition, a new multi-physical hygro-mechanical  
97 model that couples the moisture transport in concrete, drying shrinkage and  
98 plastic strain evolution will be introduced. Based on the hygro-mechanical  
99 model, an incremental form of the stress-strain constitutive relationship will  
100 be derived so that it can be implemented in general finite element (FE) rou-  
101 tine. The FE model can estimate the cracking potential under the action  
102 of diffusion and shrinkage of concrete skeleton and it can be extended to  
103 incorporate other mechanical and time-dependent (e.g. creep) action in the  
104 FE model. The model parameters were calibrated by the drying shrinkage  
105 data in this study and the crack pattern of typical configurations of building  
106 fasçade will be discussed.

## 107 **2. Experimental details**

### 108 *2.1. Materials and chemicals*

109 In this study, the cementitious material consisted of ordinary Portland  
110 cement (OPC, CEM I 52.5), calcium sulfoaluminate (CSA) cement clinker  
111 (Grade 72.5, Score Tech Mortars Co. Ltd), ground granulated blast-furnace  
112 slag (GGBS, K.Wah Construction materials Limited), magnesium expansive  
113 agent (MEA, Score Tech Mortars Co. Ltd), limestone fine (LF, Score Tech  
114 Mortars Co. Ltd) and undensified condensed silica fume (SF, Elkem Microsil-  
115 ica Grade 920U). The particle size of OPC, GGBS, MEA, CSA and LF was  
116 in similar range (in 10-100  $\mu\text{m}$ ). According to the datasheet of Elkem, the  
117 minimum specific surface area and maximum retention in 45 microns sieve  
118 of SF was 15 m<sup>2</sup>/kg and 10%, respectively. The results of elemental analysis  
119 from X-ray fluorescence (XRF) spectroscopy of the as-received raw materials  
120 are shown in Table 1.

121 The roles of limestone fine are inert filler and the nucleation site for  
122 the hydration of OPC [34] and CSA cement [27]. The water demand for  
123 complete hydration of CSA cement depends on the dosage of gypsum and it  
124 is maximum at about 30% wt while it is typical to add 15-25% wt of gypsum  
125 to CSA cement clinker [21]. The as-received CSA cement clinker was ground  
126 to particle size similar to OPC by the supplier and it was blended with 15%  
127 mass of industrial grade gypsum dihydrate ( $C\bar{S}H_2$ , Score Tech Mortars Co.  
128 Ltd) in powder form with similar particle size with OPC in a pan mixer of  
129 the laboratory before mixing with other powder. It is referred as CSA-blend  
130 in the later part of the study.

131 MEA is a common admixture for expansive cement [42]. The MEA used

132 in this study was lightly calcined industrial grade magnesium oxide. The  
133 as-received MEA was further calcined in an in-house furnace for 1 hour at  
134 800°C, 900°C and 1000°C, respectively. The purpose of further calcination  
135 was to increase the crystallinity and hence reduced the reactivity so that it  
136 could compensate the drying shrinkage in longer period.

137 To improve the workability of the mix in fresh state, high performance  
138 polycarboxylate based superplasticiser (BASF Glenium ACE 80) was used.  
139 Industrial grade boric acid (Score Tech Mortars Co. Ltd) in powder form  
140 was used to control the setting time of the mix with CSA-blend [10]. Since  
141 the mix contained all fine powder, industrial grade hydroxypropyl methylcel-  
142 lulose (HPMC, Score Tech Mortars Co. Ltd) and SF was added as viscosity  
143 modifying agent to improve the cohesiveness of the mix. The foaming agent  
144 used for the foamed concrete was fatty alcohol based liquid (BASF Rheo-  
145 cell 30), the main chemical compositions of which are 2-(2-butoxyethoxy)  
146 ethanol, dodecyl alcohol and tetradecanol mixed isomers according to the  
147 material safety data sheet from the supplier.

## 148 *2.2. Verification test*

149 Shrinkage is a characteristic material property. Any decrease of relative  
150 humidity impacts the capillary pressure, disjoining pressure and surface ten-  
151 sion in the connected porosity [40]. In nanoscopic scale (2-50 nm) of mature  
152 cementitious materials, the effect on drying shrinkage by disjoining pressure  
153 is shown to be dominant over the change in capillary pressure and surface  
154 tension [9]. The driving force of drying shrinkage ( $q_h$ ) can be estimated  
155 by considering the thermodynamical equilibrium between water vapour and  
156 liquid water given by Kelvin's law in Eq. (1) [26, 58, 9].

$$q_h = P_{gas} - P_{liq} = \frac{2\delta_w}{r_h} = -\frac{RT}{MV_f} \ln h \quad (1)$$

157 where  $P_{gas}$  is the water vapour pressure,  $P_{liq}$  is liquid water pressure,  $\delta_w$   
 158 is water surface tension,  $r_h$  is Kelvin's radius,  $R$  is gas constant,  $T$  is the  
 159 absolute temperature,  $M$  is molecular weight of water,  $V_f$  is water volume  
 160 that is equal to the volume of fully saturated pore and  $h$  is the pore relative  
 161 humidity. The capillary pressure is inversely proportional to the Kelvin's  
 162 radius of pore. According to Ziembicka [60], the drying shrinkage of cellu-  
 163 lar lightweight concrete is mainly correlated to capillary pore with pore size  
 164 ranged between 75 and 625 Å. In contrast, the bubble size distribution of  
 165 entrained bubble in foamed concrete is mainly in the range of 10-150 μm  
 166 [15], which has much less effect on drying shrinkage compared to the existing  
 167 capillary pore in the matrix. Based on the above argument, a hypothesis is  
 168 made that the drying shrinkage of non-foamed concrete is similar to foamed  
 169 concrete under the same mix formulation. This hypothesis is verified by com-  
 170 paring the drying shrinkage of 3 different mix formulations of both foamed  
 171 and non-foamed concrete. After the hypothesis is verified, only the drying  
 172 shrinkage of non-foamed concrete in normal density (about 2,000 kg/m<sup>3</sup>) is  
 173 investigated with different combinations of binder and filler without incorpo-  
 174 rating preformed foam or foaming agent.

### 176 2.3. Mix design of the verification test between foamed and non-foamed con- 177 crete

178 To compare the drying shrinkage between non-foamed concrete (normal  
 179 mortar) and foamed concrete, three sets of mix were selected as shown in Ta-

180 ble 2. The target compressive strength of foamed concrete was about 40 MPa  
181 and the wet density of foamed concrete was set at around 1,600 kg/m<sup>3</sup> [30].  
182 The first set was the blend of OPC and GGBS with mass ratio of OPC:GGBS  
183 = 2:1. The second set was the blend of OPC, GGBS and SF with mass ratio  
184 of OPC:GGBS = 1:3.2 plus 3% SF of the total mass of all powder (including  
185 itself). The third set was the blend of OPC, GGBS, SF and LF with mass  
186 ratio of OPC:GGBS = 1:3.2 plus 3% SF and 20% LF of the total mass of all  
187 powder. The water to powder ratio was fixed at 0.285. The foamed concrete  
188 samples were made by adding foaming agent of 0.15% mass of the all powder  
189 in the wet mix [17] instead of using preformed foam. The water to powder  
190 ratio, the amount of foaming agent, SP, SF and HPMC was determined by  
191 trial-and-error to achieve the rheological properties and target wet density  
192 of both non-foamed and foamed sample without segregation and excessive  
193 bleeding in the Hobart Mixer HSM 20 used in the laboratory. When foaming  
194 agent was added, the viscosity of the wet mix increased significantly. If the  
195 workability of the mix was too low, entrained bubbles by foaming were coa-  
196 lesced by excessive shear stress during mixing that reduced the compressive  
197 strength of foamed concrete significantly. It was the major reason to keep  
198 the water content of the mix similar even through it increased the water-to-  
199 binder ratio when binder was replaced by LF (N3 and F3). For each mix  
200 formulation, there were three 100 mm cubic samples for compression test  
201 and three 40 mm×40 mm×300 mm prismatic samples for drying shrinkage  
202 measurement.

203 *2.4. Mix design of non-foamed concrete*

204 After the hypothesis of using the drying shrinkage of non-foamed con-  
205 crete to approximate the foamed concrete counterpart was verified in 2.3,  
206 the compressive strength and drying shrinkage of three different strategies  
207 of mix design of non-foamed concrete was investigated. The three groups of  
208 mix design were (i) OPC-GGBS-LF blend (GI), (ii) OPC-GGBS-MEA-LF  
209 blend (GII) and (iii) OPC-GGBS-CSA-LF blend (GIII). There were 15 mixes  
210 in total and the detail mix formulations are shown in Table 3. The rationale  
211 behind each group is explained in the following.

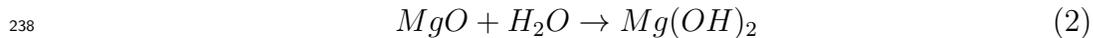
212 *2.4.1. Group I (GI), OPC-GGBS-LF blend*

213 The first group was the blend of OPC, GGBS and LF. The purpose of  
214 the first group was to reconstruct the reference of conventional composition  
215 of foamed concrete. Supplementary cementitious material GGBS was used  
216 to substitute OPC in order to reduce carbon footprint. In this study, GGBS  
217 was considered but not fly ash because the foaming agent was sensitive to  
218 unburned carbon of fly ash that might affect the foam stability from past  
219 experience of the authors. There were 5 mix formulations in GI. The effect  
220 of drying shrinkage of different mass ratios between OPC and GGBS was  
221 investigated from the first 4 mixes in group I (GI-1 to GI-4). In GI-5, part  
222 of OPC was replaced by LF to investigate the effect on drying shrinkage  
223 by reducing the binder content. In all mix in GI, 3% SF, 0.125% SP and  
224 0.005% HPMC to the total mass of all powder was added. The water to  
225 powder ratio was fixed at 0.285 as determined by the trial-and-error in the  
226 verification test.

227 *2.4.2. Group II (GII), OPC-GGBS-MEA-LF blend*

228 The second group was the blend of OPC, GGBS, MEA and LF. In the  
229 previous study, the long-term (over 20 years) drying shrinkage of concrete  
230 with 3.5-6% mass dosage of MEA and 20-30% mass of fly ash was measured  
231 [38, 42]. All concrete with MEA in [38] showed volumetric expansion and  
232 most expansion happened in the first year of the testing. The formulation of  
233 GII was based on GI-4 and 4% MEA with four different degrees of calcination  
234 to the total mass of all powder was added.

235 MEA is conventionally used as expansive agent to compensate autogenous  
236 and drying shrinkage of OPC. The hydration reaction between MEA and  
237 water to form magnesium hydroxide in Eq. (2) is expansive.

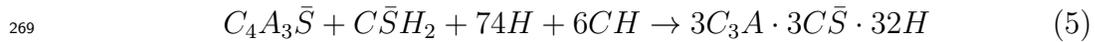
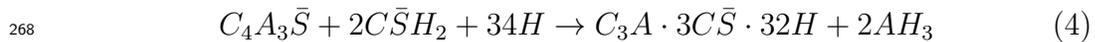


239 The reactivity of hydration of MEA depends on the concentration of  
240 surface defects of the MgO crystals. The less defects of MgO crystals, the  
241 slower the hydration rate but higher ultimate expansion [41, 42]. When  
242 magnesite is calcined under high temperature, it is firstly decomposed into  
243 MgO nanoparticles and they are sintered and coarsen to form MgO grain  
244 and the crystal defects are reduced [32]. The higher calcination temperature  
245 and longer residence time, the larger the MgO crystal size and less defect of  
246 MgO crystal that leads to lower specific area for reaction. When the defect of  
247 MgO crystal is reduced, more hydration product of MgO is forced to form on  
248 the exterior surface of the crystal. The combined effect is slower hydration  
249 rate but higher ultimate expansion.

250 *2.4.3. Group III (GIII), OPC-GGBS-CSA-LF blend*

251 The third group was the blend of OPC, GGBS, CSA-blend, and LF. All  
 252 mixes consisted of 3% SF, 0.125% SP and 0.005% HPMC of the total mass of  
 253 all powder. The water to powder ratio varied and it will be explained later.

254 CSA is used for shrinkage compensation of OPC [12, 2]. The major  
 255 phases of CSA cement clinker were ye'elimite ( $C_4A_3\bar{S}$ ) and belite ( $\alpha$ - $C_2S$ )  
 256 from the XRD pattern in Figure 3. The hydration of ye'elimite depends  
 257 on the availability of  $C\bar{S}H_2$  and calcium hydroxide ( $CH$ ) [36, 21, 45, 55].  
 258 The hydration products of pure ye'elimite are monosulfate and aluminium  
 259 hydroxide ( $AH_3$ , Eq. (3)). The reaction rate is very slow and the setting  
 260 time is usually more than 3 hours. When  $C\bar{S}H_2$  is available, the hydration  
 261 products of ye'elimite are ettringite and  $AH_3$  (Eq. (4)). While  $CH$  and  $C\bar{S}H_2$   
 262 is available, the hydration product of ye'elimite is purely ettringite (Eq. (5)).  
 263 The hydration reaction of CSA-blend in this study was dominated by Eq. (4)  
 264 and Eq. (5). If the amount of  $CH$  is not enough to hydrate all ye'elimite,  
 265 belite reacts with  $AH_3$  from either Eq. (3) or Eq. (4) to form strätlingite  
 266 ( $C_2ASH_8$ ) in Eq. (6).



271 Since the setting time of CSA-blend was as short as 15 minutes, boric  
 272 acid powder is added to retard the setting time to 60-75 minutes [10]. The

273 dosage of boric acid was determined from trial-and-error process [7] for each  
274 formulation. The dosage of boric acid in Table 3 is the ratio to the total  
275 weight of OPC and CSA-blend only. The total water content was adjusted  
276 empirically so that the mix in the fresh state maintained similar rheological  
277 properties because the effectiveness of SP to CSA-blend was different from  
278 OPC and GGBS.

279 There were six mixes in the GIII. GIII-1 and GIII-3 were the mix of CSA-  
280 blend without and with 20% LF to the total mass of all powder, respectively.  
281 GIII-2 was the mix of GGBS and CSA-blend to investigate whether CSA-  
282 blend can activate GGBS without OPC. GIII-4, GIII-5 and GIII-6 were the  
283 blend of OPC, GGBS, CSA-blend. The ratio between OPC and CSA-blend  
284 in GIII-4 was 1:1 while that of GIII-5 and GIII-6 was 1:2. 40% LF to the  
285 total mass of all powder was added in GIII-6.

## 286 *2.5. Specimens preparation and test*

### 287 *2.5.1. Sample preparation*

288 All dry powder including OPC, GGBS, CSA-blend, SF, LF, MEA, boric  
289 acid powder and HPMC were mixed at the lowest speed of Hobart Mixer HSM  
290 20 for 5 minutes. Superplasticiser (SP) was mixed with water thoroughly  
291 before adding to the mixed dry powder.

292 To fabricate foamed concrete, SP was mixed with 80% of the total water  
293 content while liquid form foaming agent was mixed with the remaining 20%  
294 water separately. The 80% water + 100% SP was thoroughly mixed with  
295 the dry powder mix. After the SP was effective, 20% water + 100% foaming  
296 agent was added to the wet mix to produce lightweight foamed concrete. The  
297 wet density of the sample was measured in 100 mm cubes.

298 For each mix, three cubic samples were prepared for wet density measure-  
299 ment and compressive strength test. In the same batch of wet mixture, three  
300 40 mm×40 mm×300 mm prisms were prepared for drying shrinkage measure-  
301 ment. A bolt was embedded at each end of the prism.

302 All samples were covered by cling wrap and cured at room temperature  
303 in the laboratory for 24 hours then the samples were demoulded. The initial  
304 length of the prismatic samples was measured by a dial gauge manually [8].  
305 Afterwards, the cubic and prismatic samples were further cured in water bath  
306 at 60°C for 7 days and at room temperature for 2 days, respectively.

### 307 *2.5.2. Drying shrinkage measurement*

308 After the prismatic samples were taken out from water bath at room  
309 temperature, they were wiped by a dry towel to remove the water on sur-  
310 face. Then, the initial length of prismatic sample was measured and it  
311 was corresponding to the day zero in the subsequent result reporting sec-  
312 tions. The samples were put in a room regulated at  $23\pm 1^\circ\text{C}$  and  $55\pm 5\%$   
313 relative humidity. The graphs of the reported drying shrinkage consisted  
314 of the average value of the three samples and the error bars. The er-  
315 ror bars were corresponding to the 90% confidence interval obtained by  
316  $\mu \pm t_{0.05,2} \cdot \sigma/\sqrt{2} = \mu \pm 2.920\sigma/\sqrt{2}$ , where  $t_{0.05,2}$  is the upper 5 percent-  
317 age point of the t-distribution with 2 degrees of freedom,  $\mu$  and  $\sigma$  are the  
318 mean and standard deviation of the three samples, respectively.

### 319 *2.5.3. Compressive strength measurement*

320 Before the compression test, all 100 mm cubic samples were air dried for  
321 another 7 days after immersing in 60°C water bath for 7 days. The reported

322 compressive strength was the average of three samples from the same batch  
 323 of mix and identical curing condition. It was corresponding to the 15th day  
 324 from sample casting. Although it was not a conventional testing condition  
 325 for compressive strength of cementitious material, it provided comparative  
 326 strength of different mixes while the compressive strength was not the main  
 327 focus in this study.

### 328 **3. Hygro-mechanical model for simulation of shrinkage**

329 This section discusses a hygro-mechanical model to simulate the diffusion,  
 330 shrinkage and plastic strain evolution of concrete. Then, the incremental  
 331 stress-strain constitutive relationship will be derived so that it can be incor-  
 332 porated in general FE model. The model parameters will be calibrated by  
 333 the drying shrinkage data in this study in section 5. Simulation examples  
 334 will be demonstrated in section 7.

#### 335 *3.1. Drying shrinkage and transport of moisture*

336 The interested  $h$  range in this study is above 50%. The main driving  
 337 force ( $q_h$ ) of drying shrinkage can be estimated by Kelvin's law in Eq. (1).

338 The resulted shrinkage strain ( $\varepsilon^{sh}$ ) can be calculated by Eq. (7) [22].

$$339 \quad \varepsilon^{sh} = \frac{\alpha_H q_h \xi}{3} \left[ \frac{1}{K} - \frac{1}{K_s} \right] \quad (7)$$

340 where  $\alpha_H$  is a constant parameter [37];  $K$  is the bulk modulus of the  
 341 porous medium;  $K_s$  is the bulk modulus of the solid skeleton;  $\xi$  is the sat-  
 342 uration factor, which can be approximated in relation to the pore relative  
 343 humidity  $h$  by Eq. (8) [37, 4],

$$344 \quad \xi = 1 - 0.75 \left[ 1 - \left( \frac{h}{0.98} \right)^3 \right] \quad (8)$$

345 The bulk moduli  $K$  and  $K_s$  can be related by Biot's coefficient  $b$  in Eq. (9).

$$346 \quad b = 1 - \frac{K}{K_s} \quad (9)$$

347 The transport of water in porous media is a diffusion-controlled process  
348 and can be described by the classic Fick's second law of diffusion [18]. If  
349 one assumes that the pore gas pressure is equal to the atmospheric pressure  
350 and the moisture capacity of concrete remains constant within the  $h$  range of  
351 50%-100%, the transport of pore moisture can be modelled in Eq. (10) [23].

$$352 \quad \frac{\partial h}{\partial t} = -\nabla \cdot (D_{eff}(h) \nabla h) \quad (10)$$

353 Note that if other types of shrinkage such as autogenous shrinkage are  
354 also considered, then the corresponding time-dependent terms shall be added  
355 to the right-hand side of Eq. (10) [23]. For uncracked concrete, the effective  
356 diffusion coefficient can be expressed in Eq. (11) [23].

$$357 \quad D_{eff}(h) = D_0 [1 + f(h)(\alpha_D - 1)] \quad (11)$$

358 where  $\alpha_D = \frac{D_1}{D_0}$ , in which  $D_0$  is the minimum of  $D_{eff}(h)$  for  $h = 0$ ;  $D_1$  is  
359 the maximum of  $D_{eff}(h)$  for  $h = 1$ ;  $f(h)$  is a hyperbolic function given by  
360 Eq. 12.

$$361 \quad f(h) = \frac{h \cdot e^{-\beta}}{1 + h(e^{-\beta} - 1)} \quad (12)$$

362 where  $\beta$  is a shape factor. The moisture flux  $\nabla h_b$ , expressed in term of  
363 the relative humidity, through the boundaries of the medium can be modelled  
364 by a convective boundary condition [23].

$$365 \quad \nabla h_b = k_h(h_b - h_{env}) \quad (13)$$

366 where  $h_b$  and  $h_{env}$  are the relative humidities of the boundary and envi-  
 367 ronment, respectively. Eq. (13) describes an imperfect moisture transfer on  
 368 the exposed surface. For perfect moisture transfer,  $h_b = h_{env}$ , as the surface  
 369 emissivity  $k_h \rightarrow \infty$ .

### 370 3.2. Mechanical responses

371 If the drying shrinkage is restrained, tensile stress is induced in concrete  
 372 and micro-cracks can develop mainly perpendicular to the gradient of the  
 373 pore humidity when the tensile strength is exceeded. The strength of con-  
 374 crete under multiaxial states of stress can be evaluated by Ottosen's four-  
 375 parameter yield criterion [44, 13].

$$376 \quad F = \alpha_p J_2 + \sigma_c(\xi_p) \left[ \lambda(\theta) \sqrt{J_2} + \beta_p I_1 \right] - \sigma_c^2(\xi_p) = 0 \quad (14)$$

377 in which the hardening parameter  $\sigma(\xi_p) = \sqrt{\varepsilon_{\mathbf{p}} \cdot \varepsilon_{\mathbf{p}}}$  is assumed to be a  
 378 function of the equivalent inelastic strain  $\varepsilon_p$ . In Eq. (14),  $\lambda(\theta)$  defines the  
 379 cross section of the yield function on the deviatoric plane and it is a function  
 380 of Lode angle  $\theta$  which can be estimated by Eq. (15).

$$381 \quad \lambda(\theta) = \begin{cases} c_p \cos \left( \frac{1}{3} \cos^{-1}(d_p \cos 3\theta) \right), & \cos 3\theta \geq 0 \\ c_p \cos \left[ \frac{\pi}{3} - \frac{1}{3} \cos^{-1}(-d_p \cos 3\theta) \right], & \cos 3\theta < 0 \end{cases} \quad (15)$$

382 The parameters  $c_p$  and  $d_p$  control the size and the shape of the cross  
 383 section, respectively. The four material parameters  $\alpha_p$ ,  $\beta_p$ ,  $c_p$  and  $d_p$  can  
 384 be calibrated from uniaxial and multiaxial strength tests. The direction of  
 385 the inelastic strain increment is described by the flow potential function, for  
 386 which the classic Drucker-Prager hyperbolic function is given by Eq. (16).

$$387 \quad G = \sqrt{J_2 + (e \cdot f_{ct} \tan \Psi)^2 + I_1 \tan \Psi} \quad (16)$$

388 where  $\Psi$  is the dilation angle measured in space at high confining pressure,  
 389  $e$  is eccentricity of the flow potential, and  $f_{ct}$  is the uniaxial tensile strength.  
 390 In Eqs. (14) and (16), the invariants of the stress tensor  $\sigma = \sigma_{ij}$  are calculated  
 391 in Eq. (17)

$$\begin{aligned}
 I_1 &= \sigma_{ii} \\
 J_2 &= \frac{1}{2} s_{ij} s_{ij} \\
 J_3 &= \frac{1}{3} s_{ij} s_{jk} s_{ki} \\
 \cos 3\theta &= \frac{3\sqrt{3}}{2} \frac{J_3}{J_2^{3/2}}
 \end{aligned} \tag{17}$$

393 In Eq. (17), the Einstein summation convention is adopted and  $s_{ij} =$   
 394  $\sigma_{ij} - \sigma_{kk} \cdot \delta_{ij} / 3$  is the deviatoric stress tensor. Since the flow potential function  
 395 in Eq. (16) is different from the yield function in Eq. (14), the inelastic strain  
 396 rate  $\dot{\varepsilon}_{\mathbf{p}}$  follows the non-associate flow rule in Eq. (18).

$$\dot{\varepsilon}_{\mathbf{p}} = \dot{\lambda} \frac{\partial G}{\partial \sigma} \tag{18}$$

398 The rate of the plastic multiplier  $\dot{\lambda}$  is determined from the Kuhn-Tucker  
 399 loading condition in Eq. (19).

$$\left\{ \begin{array}{l} \dot{\lambda} \geq 0 \\ F \leq 0 \\ \dot{F} = 0 \\ \dot{\lambda} \cdot F = 0 \end{array} \right. \tag{19}$$

401 The evolution of the yield function Eq. (14) is calibrated by the uniaxial  
 402 tensile stress-strain curve, where the post-peak stress is assumed to soften

403 exponentially after the peak stress  $f_t = k_t f_c$  in Eq. (20) [33].

$$404 \quad k_t \cdot \sigma_c(\xi_p) = \begin{cases} E_c \cdot \varepsilon_t, & \sigma_t \leq f_t \\ f_t \cdot e^{-\xi_p/\gamma_{pt}}, & \sigma_t > f_t \end{cases} \quad (20)$$

405 where  $k_t$  is the ratio of the uniaxial tensile strength to the uniaxial compressive strength;  $\xi_p = \varepsilon_t - \varepsilon_0$  is the cracking strain in which  $\varepsilon_t$  is the tensile strain  
 406 at the peak stress  $f_t$ . The effects of ageing  $t_a$  (in days) on the compressive  
 407 strength  $f_c$  and elastic modulus  $E_c$  of concrete are modelled by Eq. (21) [11].  
 408

$$409 \quad \begin{cases} f_c(t_a) & = f_{c,28} e^{s(1-\sqrt{\frac{28}{t_a}})} \\ E_c(t_a) & = E_{c,28} e^{0.5s(1-\sqrt{\frac{28}{t_a}})} \end{cases} \quad (21)$$

410 in which  $f_{c,28}$  and  $E_{c,28}$  are the strength and elastic modulus at an age  $t_a$  of  
 411 28 days;  $s$  is a coefficient which depends on the aggregate type and strength  
 412 class of cement. The area under the stress-strain curve, which depends on  
 413 the characteristic length  $\ell_{eq}$  of the element, is controlled by  $\gamma_{pt}$  defined in  
 414 Eq. (22) [33].

$$415 \quad \gamma_{pt} = \frac{G_f}{\ell_{eq} f_{ct}} - \frac{1}{2} \frac{f_{ct}}{E_c} \quad (22)$$

416 The use of the parameter  $\gamma_{pt}$  can mitigate the spurious mesh sensitivity  
 417 and ensure the energy dissipation in an element, where the crack opening is  
 418 smeared and represented by the equivalent cracking strain  $\xi_p$ , to be consistent  
 419 with the fracture energy  $G_f$ .

### 420 3.3. Incremental stress-strain relationship

421 The effects of shrinkage on the cracking potential of concrete structures  
 422 are the major interests of engineers. The behaviour of concrete structures

423 under drying can be studied by FE methods. To implement the above hygro-  
424 mechanical model in general FE routine, an incremental form of the consti-  
425 tutive relationship shall be established. Based on the series model, the total  
426 strain increment  $\Delta\boldsymbol{\varepsilon}_i$  at time step  $i$  is the linear combination of the elastic  
427 strain increment  $\Delta\boldsymbol{\varepsilon}_i^e$ , shrinkage increment  $\Delta\boldsymbol{\varepsilon}_i^{sh}$ , and instantaneous inelastic  
428 increment  $\Delta\boldsymbol{\varepsilon}_i^p$  as in Eq. (23).

$$429 \quad \Delta\boldsymbol{\varepsilon}_i = \Delta\boldsymbol{\varepsilon}_i^e + \Delta\boldsymbol{\varepsilon}_i^{sh}\mathbf{I} + \Delta\boldsymbol{\varepsilon}_i^p \quad (23)$$

430 where  $\mathbf{I} = [\delta_{ij}]$  is the  $3 \times 3$  identity matrix. The stress increment  $\Delta\boldsymbol{\sigma}_i$  is  
431 always related to the elastic strain increment  $\Delta\boldsymbol{\varepsilon}_i^e = \mathbf{D}_e^{-1} \cdot \Delta\boldsymbol{\sigma}_i$  in which  $\mathbf{D}_e$   
432 is the age-dependent elastic stiffness matrix in Eq. (24).

$$433 \quad \mathbf{D}_e = E_c(t_a) \begin{bmatrix} 1 & -\nu & -\nu & 0 & 0 & 0 \\ -\nu & 1 & -\nu & 0 & 0 & 0 \\ -\nu & -\nu & 1 & 0 & 0 & 0 \\ 0 & 0 & 0 & 2(1+\nu) & 0 & 0 \\ 0 & 0 & 0 & 0 & 2(1+\nu) & 0 \\ 0 & 0 & 0 & 0 & 0 & 2(1+\nu) \end{bmatrix} \quad (24)$$

434 in which  $\nu$  is the Poisson's ratio. By substituting  $\Delta\boldsymbol{\sigma}_i = \mathbf{D}_e \cdot \Delta\boldsymbol{\varepsilon}_i^e$  and  
435 Eqs. 14–19 into Eq. 23, then an incremental stress-strain relation is obtained  
436 in Eq. (25).

$$437 \quad \Delta\boldsymbol{\sigma}_i = \mathbf{D}_{ep} \cdot (\Delta\boldsymbol{\varepsilon}_i - \Delta\boldsymbol{\varepsilon}_i^{sh}\mathbf{I}) \quad (25)$$

438 where  $\mathbf{D}_{ep} = \mathbf{D}_e + \mathbf{D}_p$  is the incremental stiffness (Jacobian) matrix at  
439 time step  $i$ . The degradation of the material stiffness due to cracking is  
440 represented by the plastic stiffness tensor  $\mathbf{D}_p$  in Eq. (26).

$$441 \quad \mathbf{D}_p = -\frac{(\mathbf{D}_e \cdot \frac{\partial G}{\partial \boldsymbol{\sigma}}) \otimes (\frac{\partial F}{\partial \boldsymbol{\sigma}} \cdot \mathbf{D}_e)}{H + (\frac{\partial F}{\partial \boldsymbol{\sigma}} \cdot \mathbf{D}_e \cdot \frac{\partial G}{\partial \boldsymbol{\sigma}})} \quad (26)$$

442 where  $H$  is the softening parameter calculated by Eq. (27).

$$443 \quad H = -\frac{\partial F}{\partial \xi_p} \left( \frac{\partial \xi_p}{\partial \boldsymbol{\varepsilon}^p} \cdot \frac{\partial G}{\partial \boldsymbol{\sigma}} \right) \quad (27)$$

444 The coupled hygro-mechanical problem is a 2-field problem, which is de-  
445 scribed by the vector field of displacement  $\mathbf{u}$  and the scalar field of pore rel-  
446 ative humidity  $h$ . During time step  $i$ , each Gauss point or integration point  
447 of an element is provided with the increments of strain and relative humid-  
448 ity, which are interpolated from the nodal values using the prescribed shape  
449 function. In each Gauss point, the incremental constitutive equations are  
450 numerically integrated using the modified explicit Euler scheme with sub-  
451 stepping [49]. The coupled hygro-mechanical constitutive model described  
452 above was implemented in ABAQUS using user-subroutine UMAT [24].

## 453 **4. Results and discussions of laboratory experiment**

### 454 *4.1. Verification test*

#### 455 *4.1.1. Compressive strength*

456 The compressive strength of the three selected mix formulations of ver-  
457 ification test is shown in Table 2. The compressive strength of the three  
458 foamed and non-foamed concrete specimens was about 40 MPa and 90 MPa,  
459 respectively.

#### 460 *4.1.2. Drying shrinkage*

461 Figure 4 shows the micrographs of foamed concrete in the verification  
462 test. The density was about 1,500 kg/m<sup>3</sup>. The diameter of the entrained  
463 bubble was mainly between 10 and 150 micron. Figure 5a shows the drying

464 shrinkage of the non-foamed and foamed concrete. Although the compressive  
465 strength of the three selected non-foamed concrete formulations was similar,  
466 the magnitude of drying shrinkage was significantly different (varied from  
467 2,000  $\mu\epsilon$  to more than 3,000  $\mu\epsilon$ ). In Figure 5a, the trend and magnitude of  
468 drying shrinkage of foamed concrete is very close to the non-foamed coun-  
469 terparts. It is consistent to the finding in [60] that the drying shrinkage was  
470 essentially contributed by the capillary pores between 75 and 625  $\text{\AA}$  and there  
471 was little effect of the entrained bubble by foaming on drying shrinkage in all  
472 age in the test. Hence, it is justifiable to investigate the drying shrinkage of  
473 foamed concrete by measuring the drying shrinkage of non-foamed concrete  
474 counterpart.

#### 475 *4.2. Non-foamed concrete – GI*

##### 476 *4.2.1. Compressive strength*

477 For GI, the compressive strength of those mixes of the blend of OPC-  
478 GGBS was similar (between 90 and 106 MPa in Table 3) except GI-2 (about  
479 41 MPa), which did not contain any OPC.

##### 480 *4.2.2. Drying shrinkage*

481 The expansive strain of GI during water curing is shown in Table 3. The  
482 reference length of the expansive strain was taken from the length of the  
483 sample after demoulding before putting into water bath. The expansion of  
484 OPC-GGBS blends (GI-2, GI-3 and GI-4) was significantly higher (157%  
485 in average) compared with OPC mixes (GI-1 and GI-5). The results of  
486 drying shrinkage test of GI are shown in Figure 5b. The reference length  
487 of the drying shrinkage shown was corresponding to the length after water

488 curing. All drying shrinkage on the 28th day (the 31st day after casting)  
489 was generally beyond  $2,000 \mu\epsilon$  which was several times higher than normal  
490 concrete. There was little impact on drying shrinkage from replacing binder  
491 by LF (GI-1 and GI-5). The drying shrinkage of those mixes with GGBS was  
492 generally higher (GI-2, GI-3 and GI-4). When 75% mass of OPC was replaced  
493 by GGBS (GI-4), the drying shrinkage on the 28th day was increased by 50%  
494 compared with the pure OPC mix (GI-1), which was consistent to the finding  
495 in [48]. The drying shrinkage of the pure GGBS (GI-2) was lower compared  
496 to OPC-GGBS blends (GI-3 and GI-4). However, the rate of increase of  
497 drying shrinkage was higher than other mixes because the reactivity of pure  
498 GGBS in GI-2 was much slower.

#### 499 *4.3. Non-foamed concrete – GII*

##### 500 *4.3.1. Compressive strength*

501 The compressive strength of GII is higher for higher calcination temper-  
502 ature of MEA and it was all higher than the compressive strength of the  
503 reference OPC-GGBS blend in GI-4 (Table 3).

##### 504 *4.3.2. Drying shrinkage*

505 The expansive strain of GII during water curing is shown in Table 3. The  
506 expansion of the as-received MEA (GII-1) was significantly lower than (40%  
507 in average) compared with other mixes in GII. The reason may be because of  
508 the incomplete calcination of the as-received MEA. In this study, the use of  
509 MEA did not reduce drying shrinkage of the reference mix GI-4 significantly  
510 (Figure 5c). Unlike the reports from literature that with significant expan-  
511 sion with as low as 4% wt of MEA, the curing conditions of those reports

512 were completely different from this study. For example, the curing conditions  
513 were room temperature in water [38] and 20°C at 90% R.H. [19]. For GII-1,  
514 which contained 4% as-received MEA, showed the least drying shrinkage in  
515 GII. One possible explanation was that the hydration reaction of MgO in  
516 Eq. (2) was halted by the depletion of water shortly after the drying shrink-  
517 age test started. The reactivity of the as-received MEA was the fastest and  
518 the crystallinity was the lowest so that it was fast enough to compensate the  
519 drying shrinkage at early age. It can be verified from Figure 5c that the dry-  
520 ing shrinkage on the 7th day of the shrinkage test (the 10th day after casting)  
521 of GII-1 was significantly lower than the other three sets. Another possible  
522 reason of less MEA expansion observed in this study (with GGBS) compared  
523 to other studies with fly ash [38, 42] was that the pH value of the pore so-  
524 lution with GGBS was lower than fly ash because the pozzolanic reactivity  
525 of GGBS is higher. Then, the supersaturated degree of  $Mg^{2+}$  was lower in  
526 lower pH environment and it reduced the expansion near MgO particle [42].

#### 527 4.4. Non-foamed concrete – GIII

##### 528 4.4.1. Compressive strength

529 The compressive strength of GIII was significantly reduced when the mass  
530 ratio of CSA-blend to OPC was less than 2:1 (Table 3). One of the possible  
531 reasons was that part of  $C\bar{S}H_2$  reacted with tricalcium aluminate ( $C_3A$ )  
532 in OPC so that there was not enough  $C\bar{S}H_2$  to react with ye’elimite in  
533 CSA-blend and the hydration product of CSA-blend became monosulfate  
534 (Eq. (3)). When CSA-blend was blended with GGBS without OPC (GIII-5),  
535 the compressive strength was about one-third of pure CSA-blend (GIII-1).  
536 It was because the alkalinity and reactivity of GGBS was lower than OPC.

537 4.4.2. *Drying shrinkage*

538 The expansive strain of GIII during water curing is shown in Table 3. The  
539 expansion of GIII-4 was significantly lower than other mixes in GIII. It was  
540 because there was not enough gypsum for CSA hydration when OPC:CSA  
541 was 1:1. The expansion of GIII-3 was significantly higher than GIII-1. It  
542 was because of the nucleation effect of LF to accelerate the hydration of  
543 CSA [27]. The drying shrinkage of GIII was significantly lower than GI and  
544 GII (Figure 5d). GIII-1 with pure CSA-blend was the reference in GIII.  
545 The drying shrinkage on the 28th day (the 31st day after casting) of GIII-1  
546 was about 40% of the pure OPC case (GI-1). When about 75% of CSA-  
547 blend was replaced by GGBS (GIII-2), the drying shrinkage on the 28th  
548 day was about 68% of GIII-1. It was because when GGBS contacts with  
549 water, calcium hydroxide (CH) was released and the hydration reaction of  
550 CSA-blend was changed from Eq. (4) to Eq. (5), which consumed much more  
551 water for hydration and hence less free water was left for drying shrinkage.  
552 Although GGBS consumed CH through pozzolanic reaction, it happened only  
553 after 2 to 3 days [6] while all  $C\bar{S}H_2$  was consumed in 48 hours according to the  
554 XRD result in [21]. So, pozzolanic reaction of GGBS followed the complete  
555 hydration reaction of CSA-blend in Eq. (5). When about 20% of CSA-blend  
556 was replaced by LF (GIII-3), the drying shrinkage increased dramatically.  
557 When CSA-blend was replaced by LF and there was no CH provided, the  
558 water consumption during hydration of CSA-blend was less (Eq. (4)) and  
559 there was more free water left in the mix and contributed higher drying  
560 shrinkage.

561 From the previous argument, CH was beneficial to reduce the drying

562 shrinkage of the blend of CSA-blend that the drying shrinkage should be  
563 lower by using OPC. However, when OPC-CSA-blend ratio was kept at 1:1  
564 (GIII-4) and 20% LF of the total mass of all powder, the drying shrinkage  
565 was about 11% higher than GIII-1. Although OPC provided CH for CSA-  
566 blend to form denser matrix, the additional  $C\bar{S}H_2$  was consumed by OPC  
567 to react with  $C_3A$  and monosulfate to form ettringite. Hence, there was not  
568 enough  $C\bar{S}H_2$  for CSA-blend and the hydration of CSA-blend became Eq. (3)  
569 and it could be verified by observing the compressive strength of GIII-4 is  
570 lower than GIII-1 and the micrographs in Figure 6 of foamed sample. There  
571 were needle-like crystals formed in GIII-5 (Figure 6b) compared with GIII-4  
572 (Figure 6a). Although there is no further characterisation of the crystals,  
573 it may be ettringite by comparing them with the micrographs in [50]. In  
574 addition, it resulted higher amount of free water remained in GIII-4 than  
575 GIII-1 because the water demand of Eq. 3 of GIII-4 was less than GIII-1 from  
576 Eq. 4 with the same given water content. However, LF acted as nucleation  
577 site to accelerate the hydration of CSA-blend that explained the observation  
578 of the trends of drying shrinkage of GIII-1 and GIII-4 are similar. When the  
579 OPC to CSA-blend ratio was increased to 1:2 (GIII-5), the drying shrinkage  
580 was significantly reduced compared to all other mix in GIII. The reason was  
581 that the amount of  $C\bar{S}H_2$  from CSA-blend was enough for both hydration of  
582 ye'elimite in CSA clinker and  $C_3A$  of OPC. Hence, the free water remained  
583 in the matrix was the least in all GIII mixes.

584 There was 40% LF to the total mass of all powder in GIII-6. However,  
585 the drying shrinkage of GIII-6 was much higher than GIII-5. The addition  
586 of LF does not change the water demand of the reaction significantly so the

587 free water remained in the mix was higher and hence it resulted of higher  
588 drying shrinkage.

589 In summary, the key factors to determine the magnitude of drying shrink-  
590 age in GIII are (i) the free water content and (ii) the hydration reaction  
591 (Eq. (4), Eq. (5) or Eq. (3)), which determines the total water consumed in  
592 hydration.

#### 593 4.5. Summary of the drying shrinkage test of GI, GII and GIII

594 Although the drying shrinkage approaches asymptotic value in long term,  
595 to compare the relationship of expansion during the curing stage, rate of  
596 drying shrinkage and the drying shrinkage at the 28th day after the test, the  
597 dry shrinkage versus time graph are plotted in semi-log scale of time, the  
598 relationship can be approximated by a linear line (Figure 7). Table 4 shows  
599 the best-fitted coefficients and the  $R^2$  by using the least square method of  
600 the experimental drying shrinkage data in Eq. (28).

$$601 \quad \varepsilon^{sh}(t) = a \ln \left( \frac{t}{28} \right) + c \quad (28)$$

602 where  $a$  and  $c$  are constants,  $t$  is time in days and  $\varepsilon^{sh}$  is the shrinkage  
603 strain.  $c$  is the drying shrinkage on the 28th day (the 31st day from casting)  
604 and  $a$  is the exponent that indicates the rate of increase of drying shrinkage.  
605 All data shows  $R^2 > 0.9$  except GIII-5 ( $R^2 = 0.898$ ). In general, the drying  
606 shrinkage on the 28th day of GIII is about half of GI and GII. Also, the rate  
607 of increase of drying shrinkage of GIII, in general, was about half of GI and  
608 GII. Hence, the use of CSA-blend to control drying shrinkage is an effective  
609 approach and it is consistent to the finding in [2].

610 To compare the expansive strain during curing, the rate of drying shrink-  
611 age ( $a$ ) and the 28th day drying shrinkage ( $c$ ) in Table 4, There is no strong  
612 relationship observed between the initial expansion during the curing stage  
613 and drying shrinkage in testing stage.

## 614 5. Calibration of the model parameters

615 The materials with the lowest drying shrinkage in each group: GI-5, GII-1  
616 and GIII-5 are further investigated for their feasibility in developing full-scale  
617 foamed concrete member, on which the time variation of drying shrinkage is  
618 simulated using the coupled hygro-mechanical FE models and compared with  
619 the experimental results in section 4. The four parameters of the Ottosen  
620 yield criterion can be calibrated by the following strength data at an age  
621  $t_a$  of 28 days: uniaxial compressive strength  $f_{c,28}$ , uniaxial tensile strength  
622  $f_{t,28} = k_t f_{c,28}$ , equal biaxial compressive strength  $f_{bc,28} = 1.16 f_{c,28}$ , and the  
623 triaxial stress states on the compressive and tensile meridians. Following the  
624 approach of Ottosen [44], the parameters for each material are adjusted using  
625 the least square method to give the best fits of both compressive and tensile  
626 meridians. Figure 8a shows the comparison of the triaxial test data (after  
627 Ottosen [44]) with the fitted Ottosen yield criterion in the meridian planes for  
628 foamed GI-5. The effects of aging on the strength and stiffness are modelled  
629 by Eq. (21), where the coefficient  $s$  is taken as 0.2. The yield surface evolves  
630 with the equivalent plastic strain  $\xi_p$ , which the rate of strength degradation as  
631 described by Eq. (20) is governed by the fracture energy  $G_f$ . The simulated  
632 uniaxial tensile stress-displacement curves are plotted in Figure 8b. The  
633 calibrated material elastoplastic parameters for GI-5, GII-1 and GIII-5 are

634 provided in Table 5.

635 The material parameters for the drying shrinkage model, as shown in  
636 Table 6, are calibrated using the data from the drying shrinkage test results  
637 in section 4. The values of the gas constant and molar volume of water  
638 at the room temperature are given as  $8.31 \text{ J}\cdot\text{mol}^{-1}\cdot\text{K}^{-1}$  and  $18 \text{ cm}^3\cdot\text{mol}^{-1}$ ,  
639 respectively. The models are meshed using 8-node linear solid elements as  
640 shown in Figure 9a. The boundary conditions of ambient temperature and  
641 relative humidity are defined in Figure 9b to reflect the actual test conditions.  
642 Good agreements between the simulated drying shrinkage curves and the test  
643 results can be seen in Figure 10a.

644 The cross-sectional shrinkage distributions due to drying are also captured  
645 by the models. Figures 10b-d show the simulated evolutions of shrinkage  
646 distributions across the mid-section C-C as marked in Figure 9. The drying-  
647 shrinkage always has the maximum values on the surfaces and its values  
648 gradually decrease with the depth from the surface until reaching its mini-  
649 mum values on the mid-planes. When the internal moisture gradually diffuses  
650 to the surfaces and evaporates to the environment, equilibrium of the inter-  
651 nal relative humidity will be eventually established and the cross-sectional  
652 shrinkage distribution will have uniform values. The internal relative humid-  
653 ity distribution of GI-5 and GII-1 achieves the equilibriums in shorter time  
654 as compared with GIII-5, since the effective diffusion coefficient ( $D_0$  and  $D_1$ )  
655 of GIII-5 is much lower. As a result, the cross-sectional shrinkage distribu-  
656 tion in GIII-5 also takes longer time to reach the uniform value (Figure 10).  
657 Nevertheless, the actual cross-sectional shrinkage distributions could be com-  
658 plicated by the internal creep. Self-equilibrium internal stresses can develop

659 in the specimens due to the material inhomogeneities and nonuniform strain  
660 distribution. Hence, even without the application of external loading, the in-  
661 ternal stresses can induce internal creep, which can be coupled with the free  
662 shrinkage as given by Figure 10b-d. Yet, this secondary effect is normally  
663 relatively minor and hence ignored by most shrinkage models for concrete  
664 (e.g. [23]).

## 665 **6. Scaled-up in field test**

666 GIII-5 was scaled-up in a full scale twin-screw mixer of a concrete produc-  
667 tion plant. In the previous laboratory study, the rheology was determined  
668 empirically so that the target density in the range of 1500-1700 kg/m<sup>3</sup> could  
669 be fabricated without segregation and excessive bleeding consistently. Also,  
670 the setting time was tuned to be around 60-75 minutes. The main objective  
671 was to verify the scalability of the low drying shrinkage formulation devel-  
672 oped in laboratory. A full scale reinforced foamed concrete slab was made  
673 (Figure 11). The dimensions of the slab were 2.9 m×2.5 m×0.15 m. There  
674 were two layers of T10 steel reinforcing mesh with 250 mm centre-to-centre  
675 spacing. The concrete cover was 30 mm.

676 The foamed concrete was made by preformed foam. The targeted wet  
677 density was 1,600 kg/m<sup>3</sup>. After mixing all dry powder in the mixer for 1  
678 minute, water and superplasticiser was added. The wet mix was continu-  
679 ously mixed for another 1 minute. The wet mix was poured into a concrete  
680 truck for continuous mixing. The preformed foam was pumped from the  
681 output of a home-made foam generator from the concrete producer directly  
682 into the concrete truck barrel. The amount of foam added was determined

683 by the targeted density and the duration of foam pumping. The rate of  
684 foam generated was calibrated in terms of flow rate ( $\text{m}^3/\text{s}$ ). The foamed  
685 concrete was poured into 100 mm cubes to measure the wet density imme-  
686 diately on-site. The averaged wet density of three samples was  $1720 \text{ kg}/\text{m}^3$ .  
687 The averaged 28th day compressive strength of three cubic samples cured  
688 in standard condition was 29.2 MPa. Three  $40 \text{ mm} \times 40 \text{ mm} \times 300 \text{ mm}$  prisms  
689 were cast for drying shrinkage measurement. There were three more sets  
690 of prisms without foaming based on the common strategies of the concrete  
691 producer to reduce drying shrinkage by using gypsum dihydrate ( $C\bar{S}H_2$ ) and  
692 gypsum anhydrite ( $C\bar{S}$ ). The binder (OPC, CSA-blend and GGBS) of GIII-  
693 5 was replaced by (i) 90% OPC + 10%  $C\bar{S}H_2$ , (ii) 90% OPC + 10%  $C\bar{S}$   
694 and (iii) 90% OPC + 5%  $C\bar{S}$  + 5%  $C\bar{S}H_2$ . There was 20% LF of the total  
695 mass of all powder. The dosage of SP, SF and HPMC was the same as the  
696 previous study. The water to powder ratio was kept at 0.285.

697 The slab was exposed in ambient semi-outdoor environment with canopy  
698 to prevent direct exposure to rainfall and sun radiation without temperature  
699 regulation. The location of field test was in Shunde, Guangdong province of  
700 China. The slab was fabricated in late January 2016. In the 193-day of field  
701 test, the average, minimum and maximum temperature was  $18^\circ\text{C}$ ,  $0^\circ\text{C}$  and  
702  $37^\circ\text{C}$ , respectively. After 193 days of field test, there was no visible suspicious  
703 drying shrinkage crack. The 4 sets of prisms were exposed to the same  
704 environment with the slab. The summary of the drying shrinkage after 193  
705 days from casting in the field test is shown in Table 7. The drying shrinkage  
706 of foamed GIII-5 was much lower compared to the common strategies of  
707 reducing drying shrinkage of the concrete producer and it was much smaller

708 in actual ambient environment than the more severer artificial environmental  
709 in laboratory.

## 710 **7. Simulation of full scale panels by FE model with hygro-mechanical** 711 **model**

### 712 *7.1. Model development*

713 The potential crack developments of three prototypes of reinforced con-  
714 crete panels (Figure 12), made of GI-5, GII-1 or GIII-5, are studied by FE  
715 simulations using the calibrated hygro-mechanical models in section 5. The  
716 panels have the same height and thickness of 2.9 m and 0.15 m, respectively.  
717 The horizontal widths of Type I, Type II and Type III panels are 1 m, 2 m  
718 and 2 m respectively. Type III panel has a window-opening with sizes of  
719 1.2 m × 1.5 m. There are two layers of reinforcement mesh of T10 steel with  
720 approximate 250 mm centre-to-centre spacing. The material models and  
721 meshing for concrete are similar to those in section 5. The reinforcement  
722 mesh is modelled by truss elements with the same meshing size as the solid  
723 elements of concrete. The steel reinforcement is assumed to be impermeable  
724 to moisture. The elastic modulus and Poisson's ratio of steel is 200 GPa and  
725 0.3, respectively. The environmental exposure conditions of the panels are  
726 defined to be constant relative humidity of 55% and temperature of 23°C for  
727 90 days.

### 728 *7.2. Simulation results and discussion*

729 As an illustration, the 90th day crack pattern, principal strain directions  
730 and von Mises distribution in steel reinforcement mesh the GI-5 panel on the

731 mid-plane are shown in Figure 13. The equivalent crack width  $\omega$  is calculated  
 732 by Eq. (29).

$$733 \quad \omega = \ell_{eq} \cdot \varepsilon_1^p \quad (29)$$

734 where  $\varepsilon_1^p$  is the major principal inelastic strain. In the developed ABAQUS  
 735 subroutine, the equivalent crack width at each integration point is stored as  
 736 the state variable SDV19.

737 Figure 13a shows the contour plots of the equivalent crack width distri-  
 738 butions of the three panels. The corresponding crack opening directions are  
 739 depicted by the maximum principal strain directions as shown in Figure 13b.  
 740 The corner regions of the panels have the fastest rate of moisture loss and the  
 741 resulted drying shrinkage is also the largest in those regions. The shrinkage  
 742 is then restrained by the steel reinforcement and tensile stress will continue  
 743 to develop until the tensile strength of concrete is exceeded. As a result,  
 744 the major cracks start to propagate from near the corners to the mid-points  
 745 between two adjacent corners, forming a closed loop of crack path denoted as  
 746 the loop C as shown in Figure 13. For Type III panel with a window opening,  
 747 another closed crack path is formed around the window opening but with the  
 748 smaller crack width compared with that of the outer crack path. As shown  
 749 in Figure 13, the stress in the reinforcement near the cracked regions is also  
 750 lower than that in the inner regions with minor or no cracking. Besides the  
 751 maximum crack width, the average crack width along the major crack path  
 752 C is evaluated in Eq. (30).

$$753 \quad \omega_c = \frac{1}{L_c} \oint_c \omega \cdot dl \quad (30)$$

754 where  $L_c$  is the perimeter of the loop C.

755 Evolutions of the maximum crack widths near the corner regions and the  
756 average crack width along the perimeter of the loop C are plotted against  
757 the days of curing as shown in Figure 14. It can be seen that the maxi-  
758 mum and average crack widths developed in all three types of panel have  
759 similar values, but the Type II panel has the largest cracks on the 90th day.  
760 The maximum crack widths developed in the Type II panel after 90 days  
761 are 0.1579 mm, 0.2117 mm, and 0.0666 mm for GI-5, GII-1, and GIII-5 re-  
762 spectively, and the corresponding average crack widths on the 90th day are  
763 0.1275 mm, 0.1721 mm, and 0.054 mm respectively. The regression relation-  
764 ships between the free shrinkage and crack width development are shown  
765 in Figure 14d. Although the shrinkage crack opening from simulation is  
766 smaller than typical threshold of 0.3 mm, it may be exacerbated by thermal  
767 and mechanical effects which have not been considered in the simulations but  
768 the FE model can be extended to incorporate the mechanical, thermal and  
769 time dependent effect on the crack potential based on the hygro-mechanical  
770 model and incremental stress-strain constitutive relationship. Furthermore,  
771 early corrosion of the reinforcement in the RC panels can occur if the crack  
772 width reaches more than 0.2 mm after 90 days of curing. Therefore, GIII-5,  
773 which has the smallest crack width of less than 0.1 mm in the simulations, is  
774 suggested for the fabrication of foamed concrete façades.

## 775 8. Conclusions

776 In this paper, the benefits of using lightweight prefabricated permanent  
777 formwork was discussed. The verification test showed that the drying shrink-  
778 age of foamed concrete in the density range around 1,500 kg/m<sup>3</sup> could be

779 approximated by the non-foamed counterpart. Three different strategies of  
780 reducing drying shrinkage were investigated. The OPC-GGBS blend and  
781 MEA showed excessive drying shrinkage ( $>2,000 \mu\epsilon$ ). The CSA-blend could  
782 significantly reduce the drying shrinkage. The formulation developed in lab-  
783 oratory was successfully verified by fabricating a full scale reinforced foamed  
784 concrete in a field test. The drying shrinkage of the developed formula-  
785 tion in the field test was significantly smaller than the standard environment  
786 in laboratory. It was also superior to the conventional strategies of concrete  
787 producer by using anhydrite or dihydrate to reduce drying shrinkage of OPC.

788 A hygro-mechanical model was developed to consider the diffusion of  
789 moisture, shrinkage and plastic strain evolution. The incremental stress-  
790 strain constitutive relationship of the model was derived and it could be  
791 incorporated in general FE routine. The model was calibrated by the results  
792 of drying shrinkage test in this study. The simulation demonstrated how  
793 cracking potential could be examined and it could help the engineers to  
794 model the crack formation potential with the consideration of mechanical,  
795 thermal, time-dependent and other possible factors in general FE routine.

## 796 **9. Acknowledgement**

797 The work of this paper is sponsored by CC003/12 from Nano and Ad-  
798 vanced Materials Institute Limited, Hong Kong and Global Challenges Re-  
799 search Fund of Engineering and Physical Sciences Research Council, UK  
800 (EP/P510749/1/R33466/R33471). Thank you for the technical support from  
801 the Housing Department of the government of Hong Kong SAR, Gammon  
802 Construction Limited and Quon Hing Concrete Company Limited.

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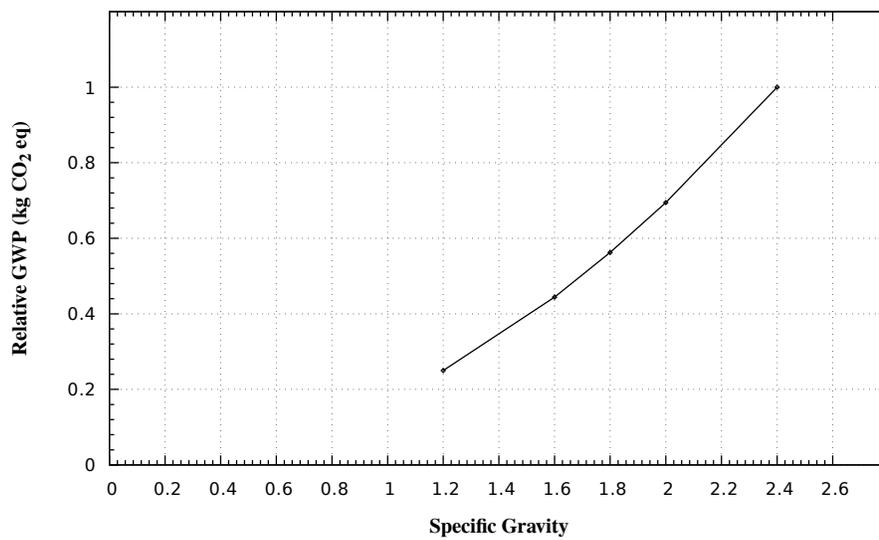
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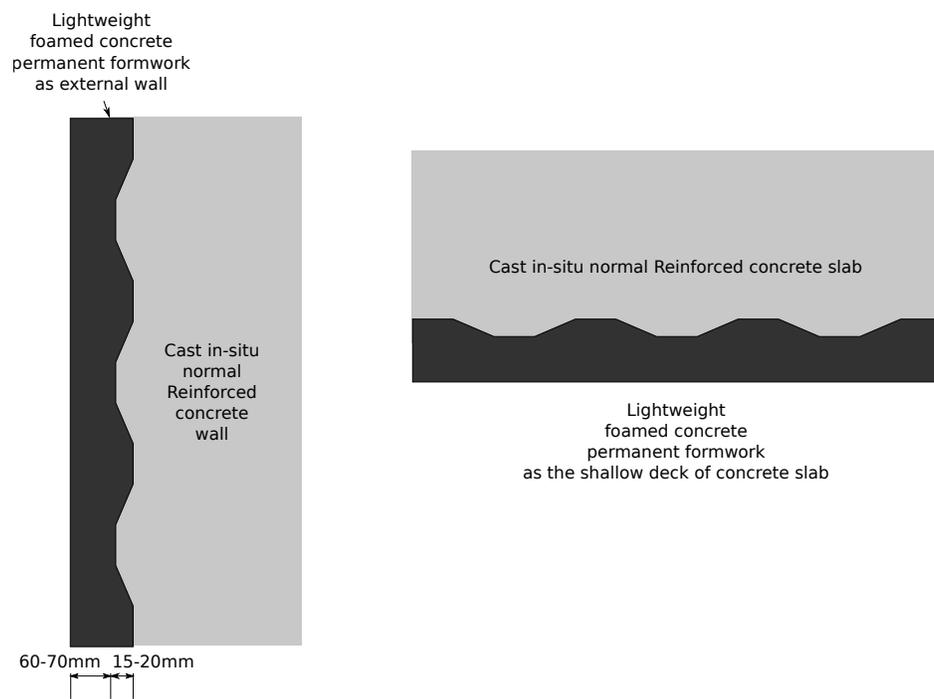
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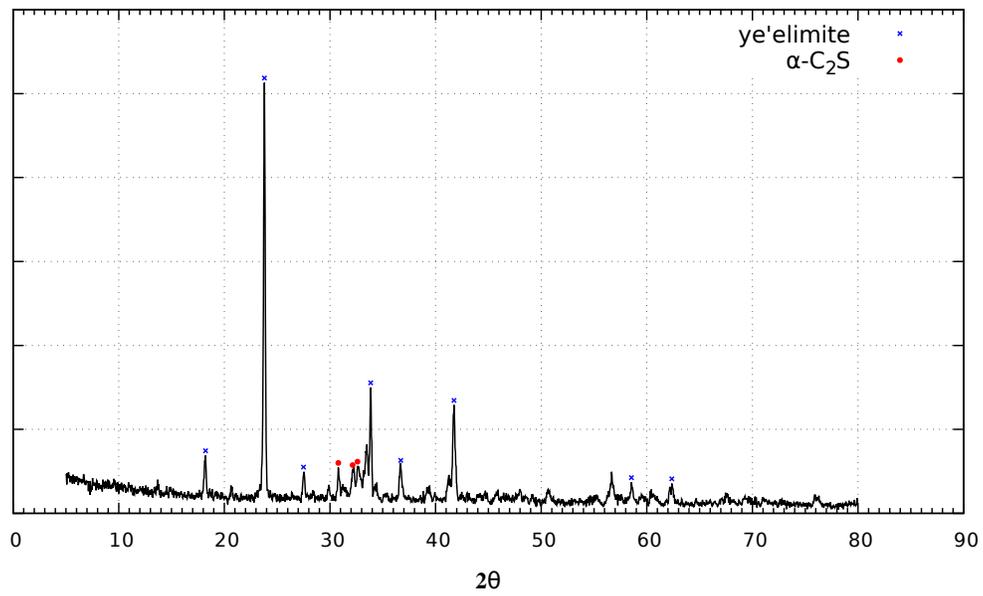
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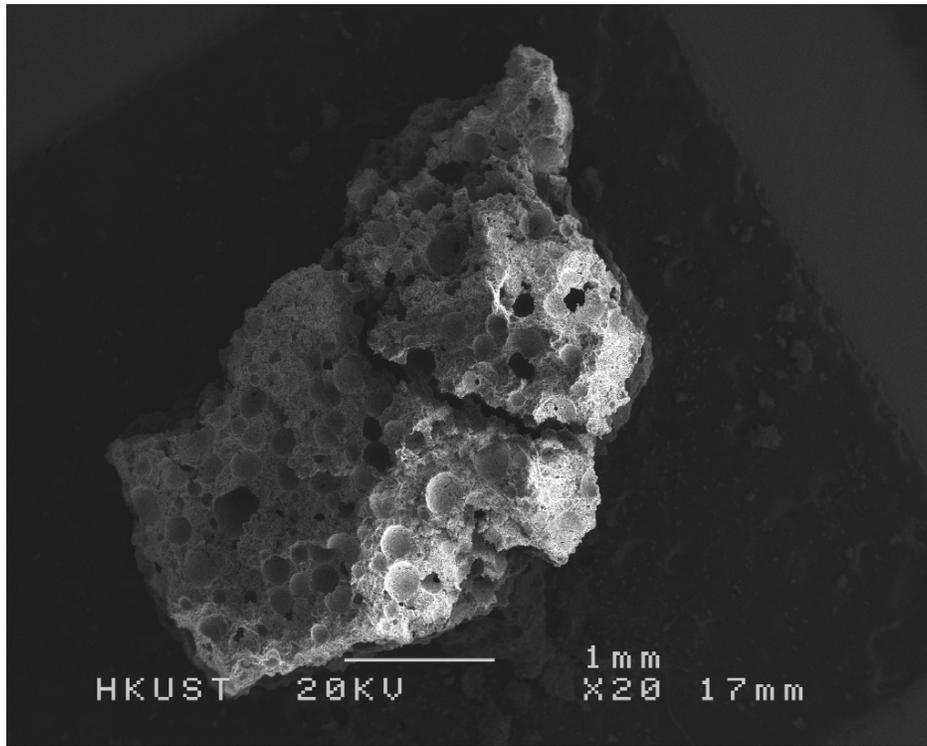
**Figure 1:** Relative global warming potential (GWP) by lorry transport for 200 km with different density of 12 full scale building façades.



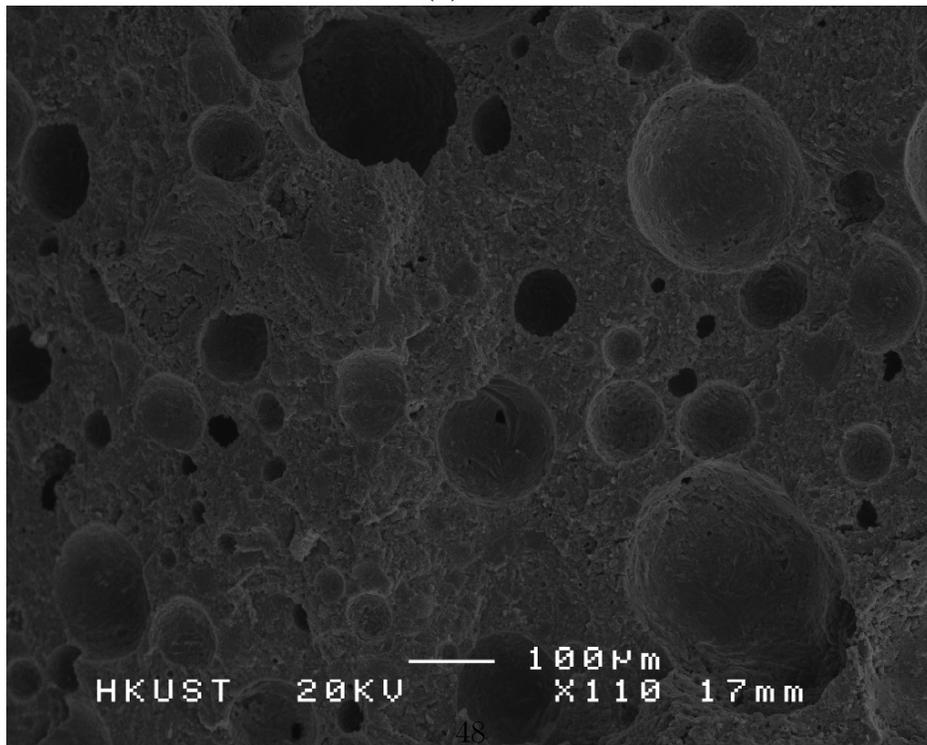
**Figure 2:** Typical sandwich structure of prefabricated reinforced concrete panel with enhanced thermal insulation. (a) structural wall and (b) floor slab.



**Figure 3:** XRD pattern of the as-received CSA cement clinker.

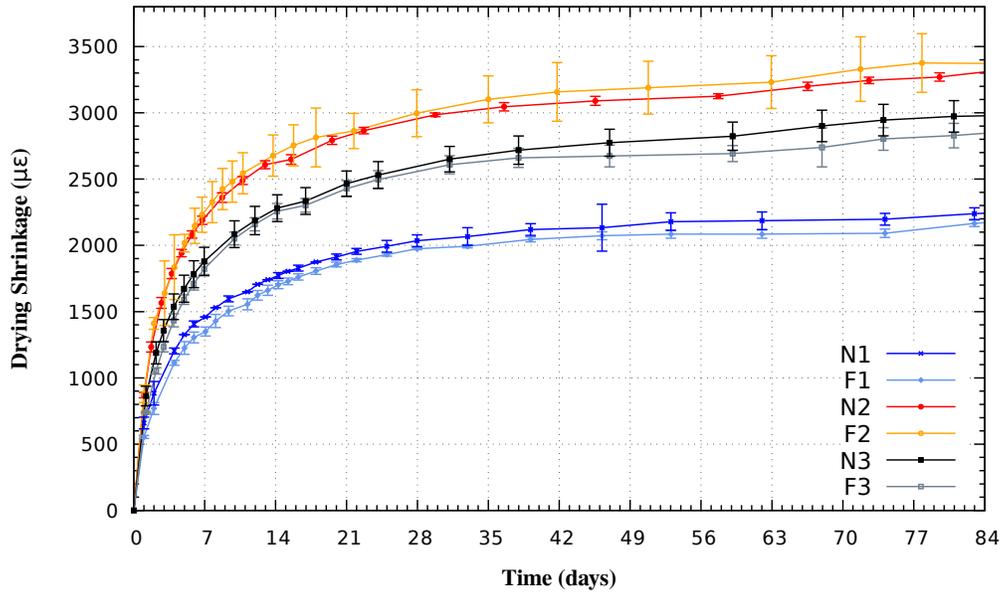


(a) 20X

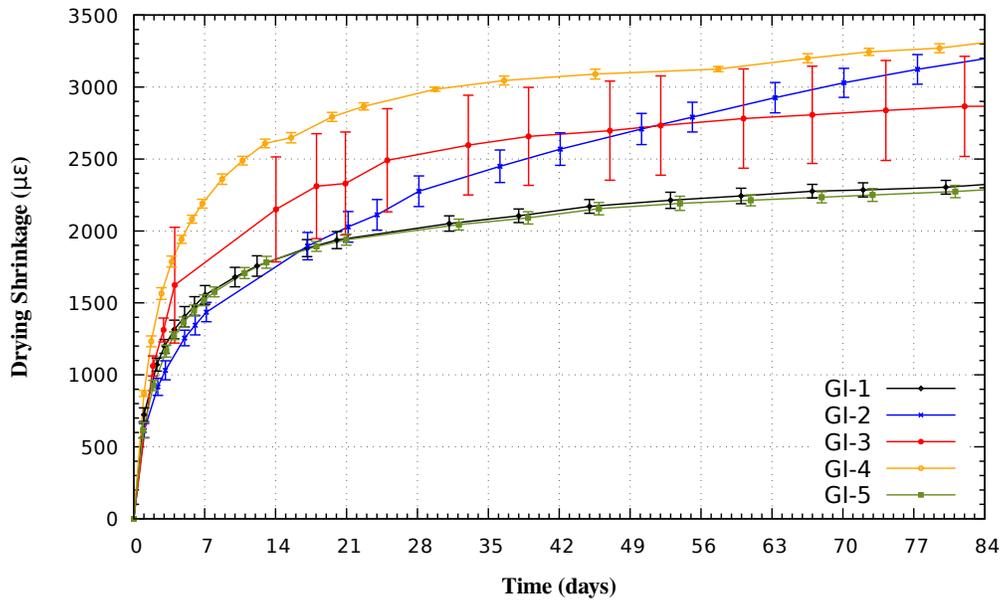


(b) 110X

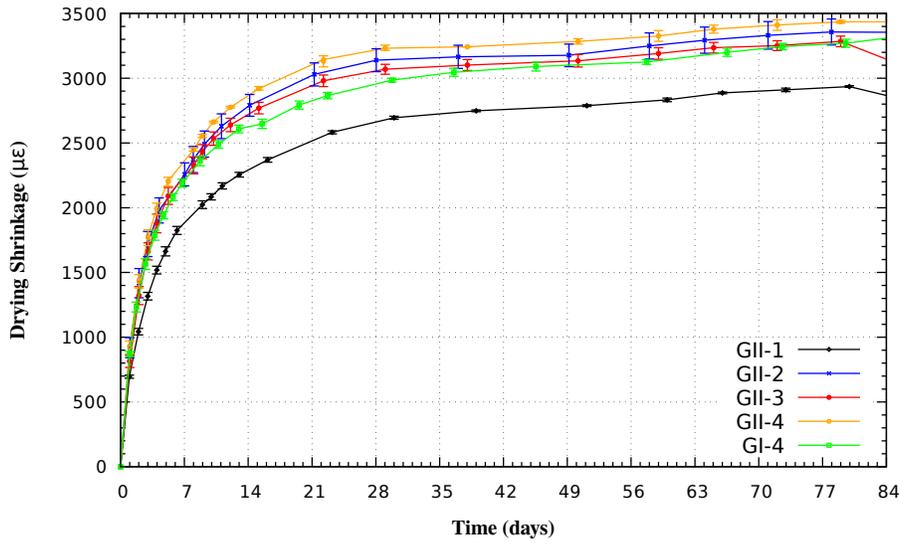
**Figure 4:** Micrograph of foamed concrete of wet density about  $1,500 \text{ kg/m}^3$ .



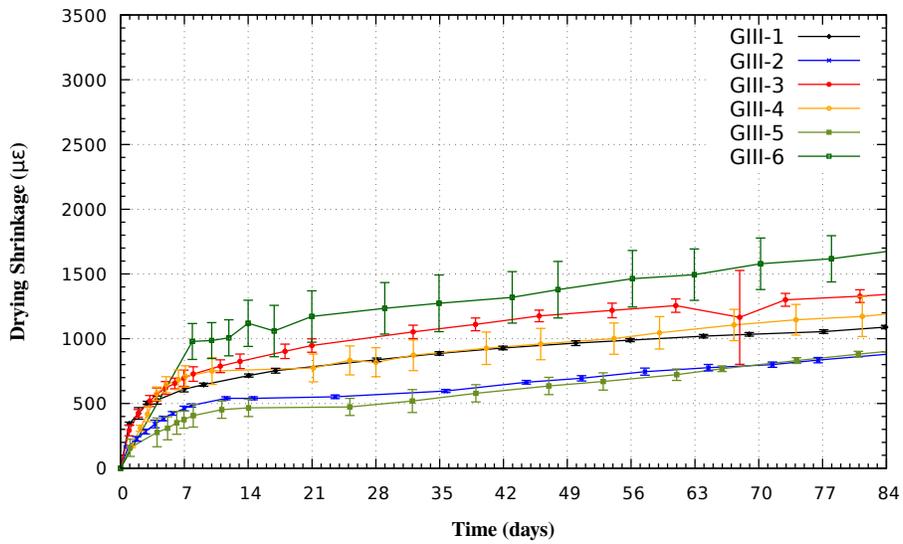
(a) Verification test



(b) GI

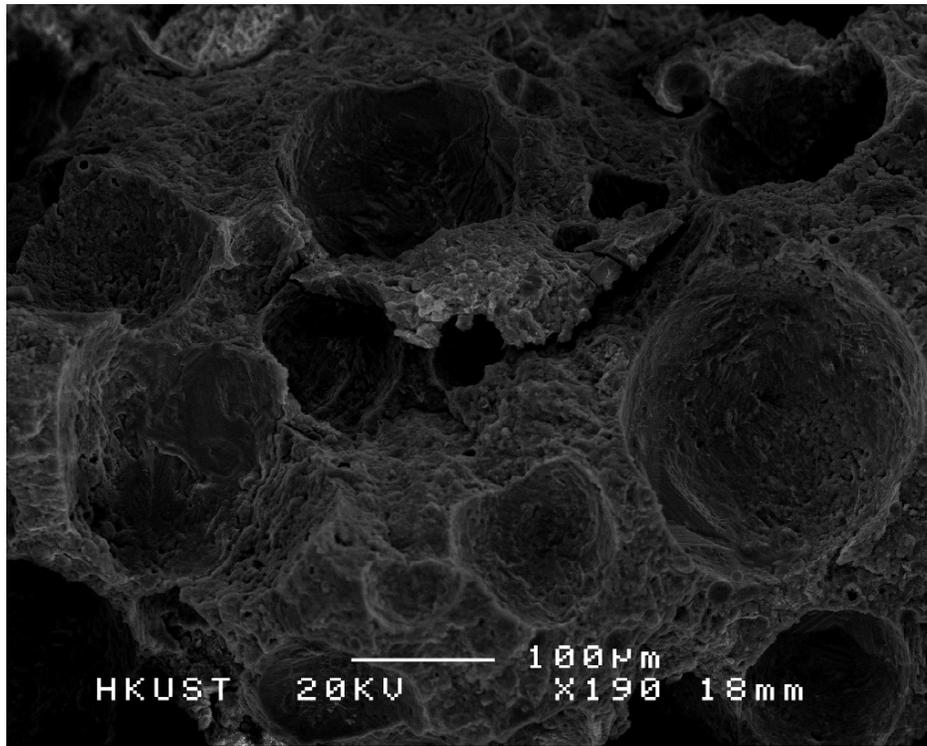


(c) GII

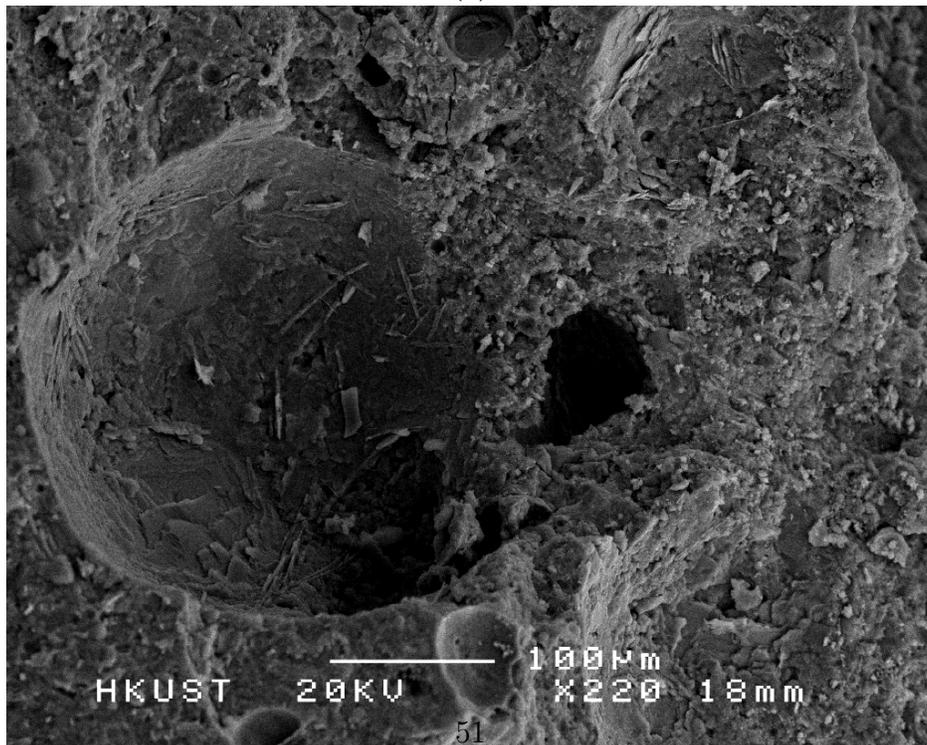


(d) GIII

Figure 5: Results of drying shrinkage test.

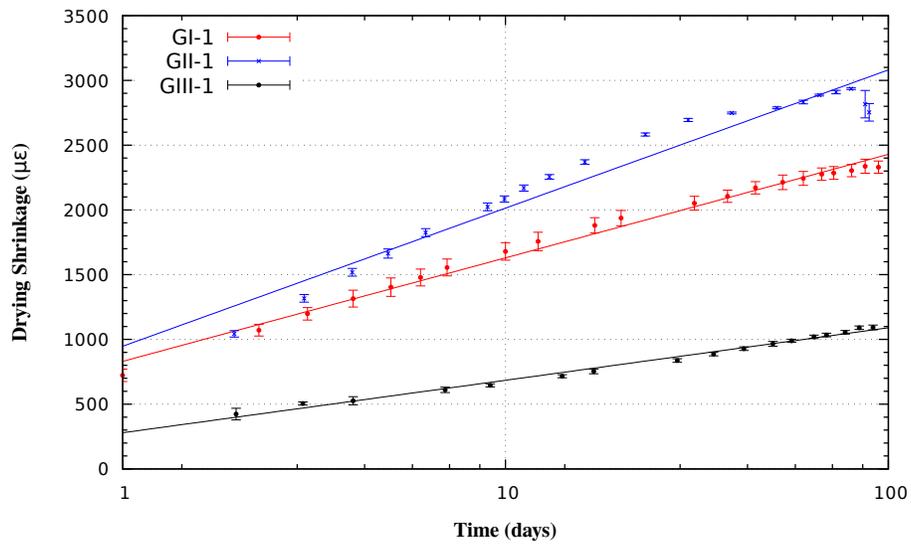


(a)

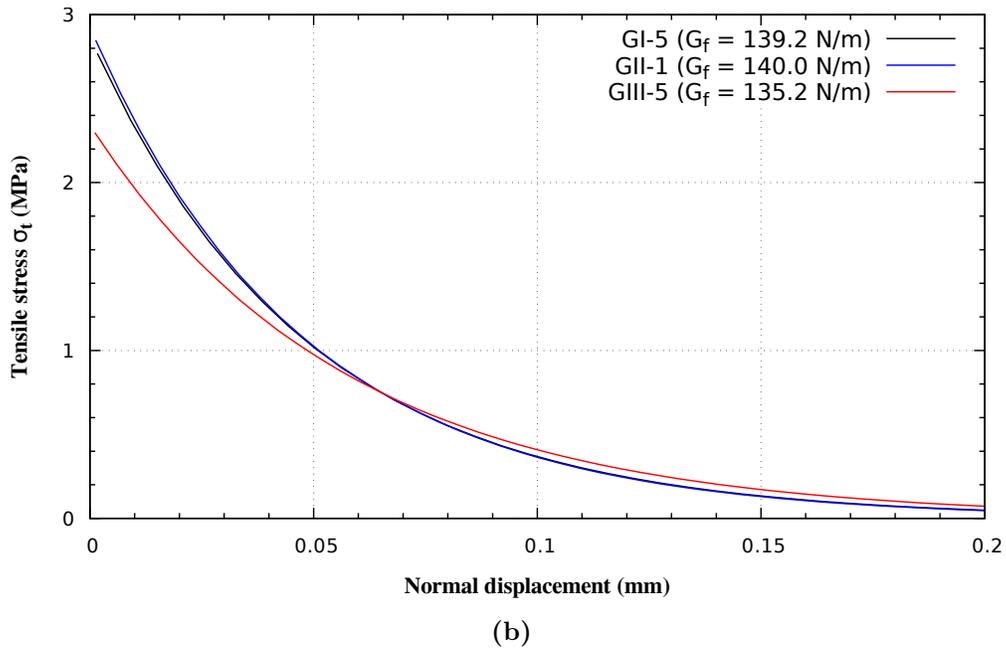
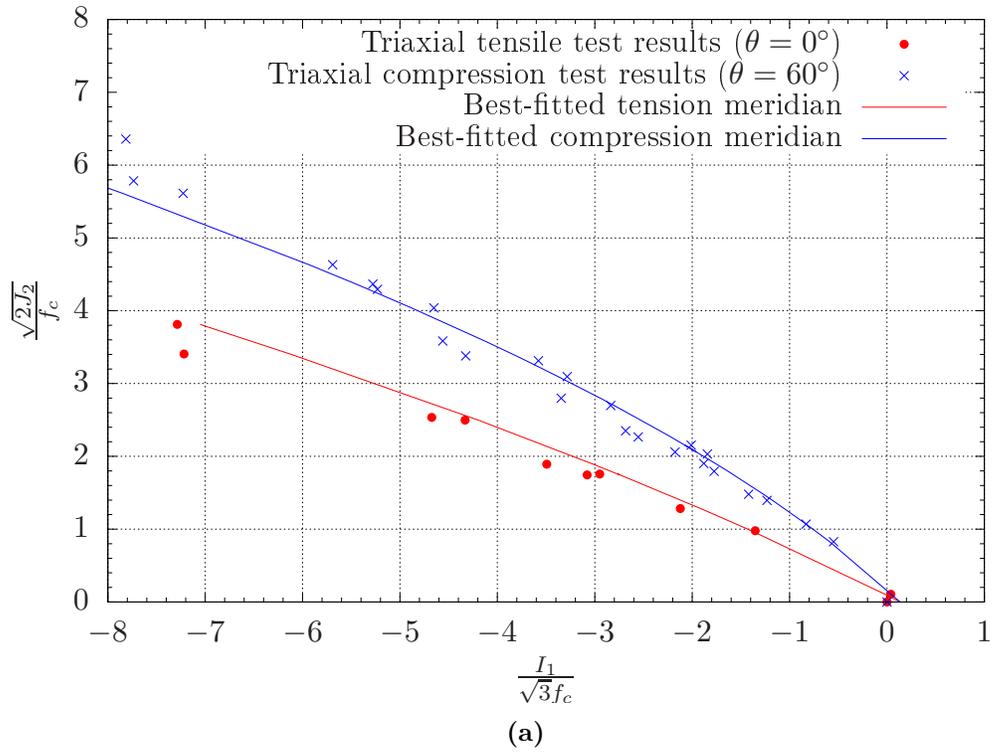


(b)

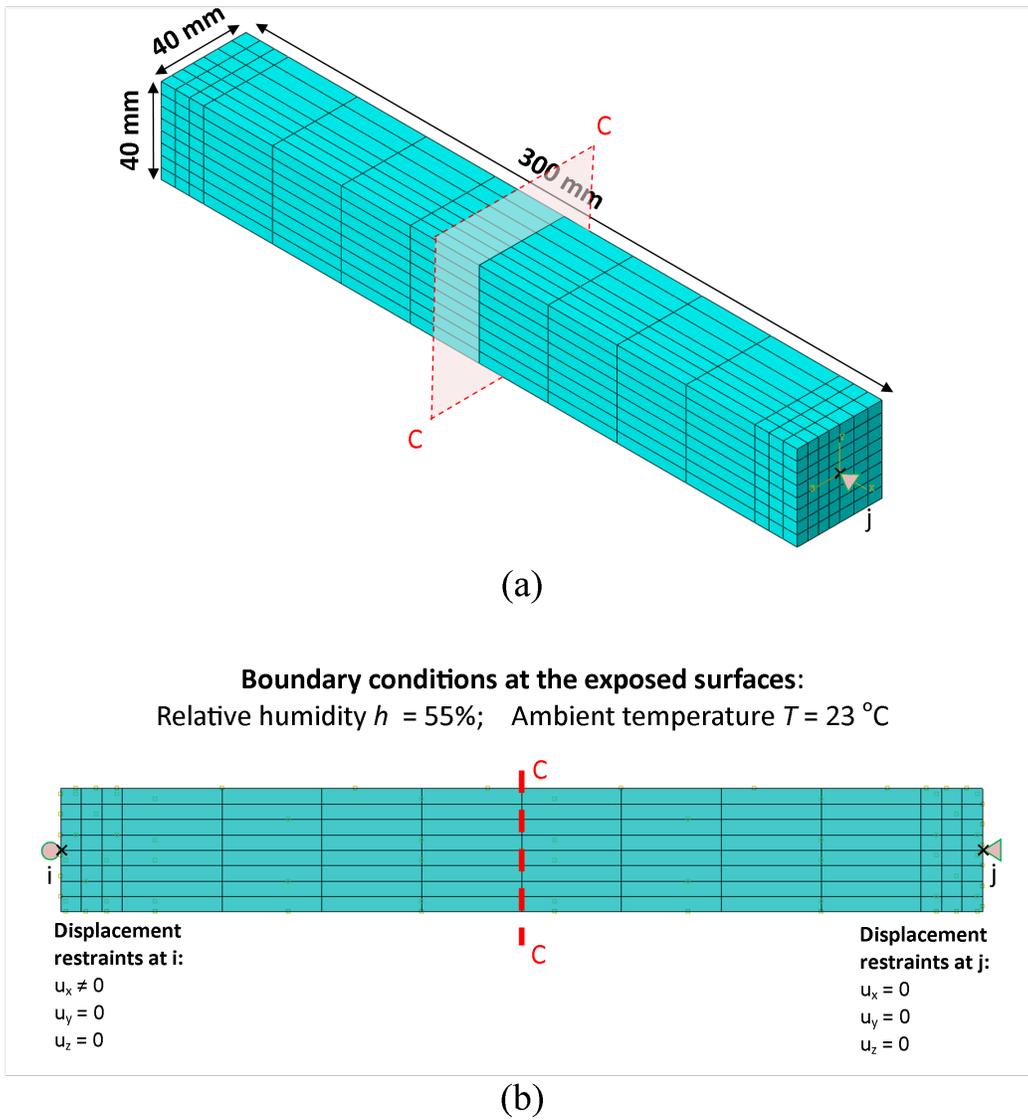
**Figure 6:** Micrographs of (a) foamed GIII-4 and (b) foamed GIII-5.



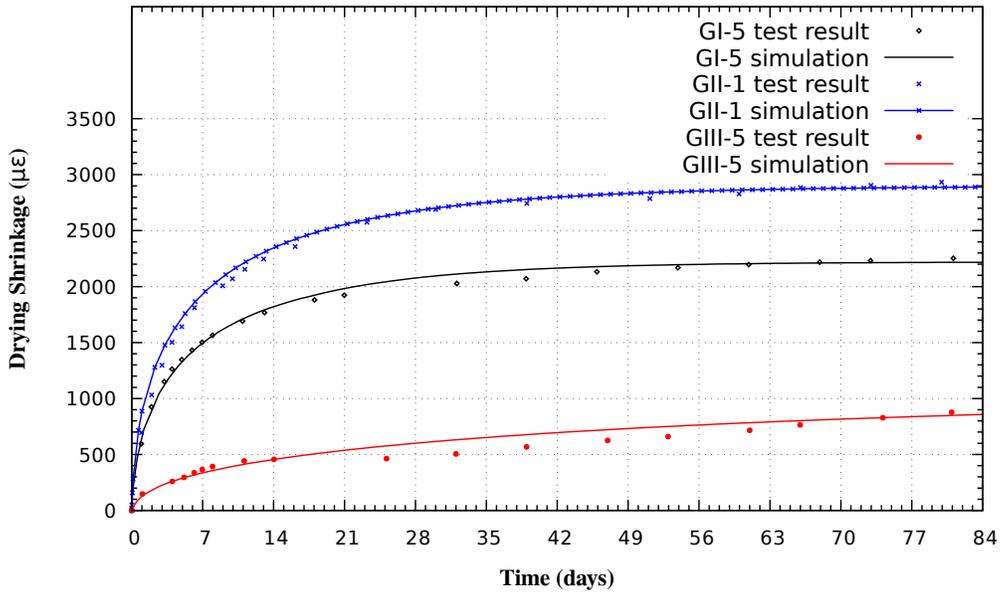
**Figure 7:** Semi-log plot of typical results of drying shrinkage test and the best-fitted lines.



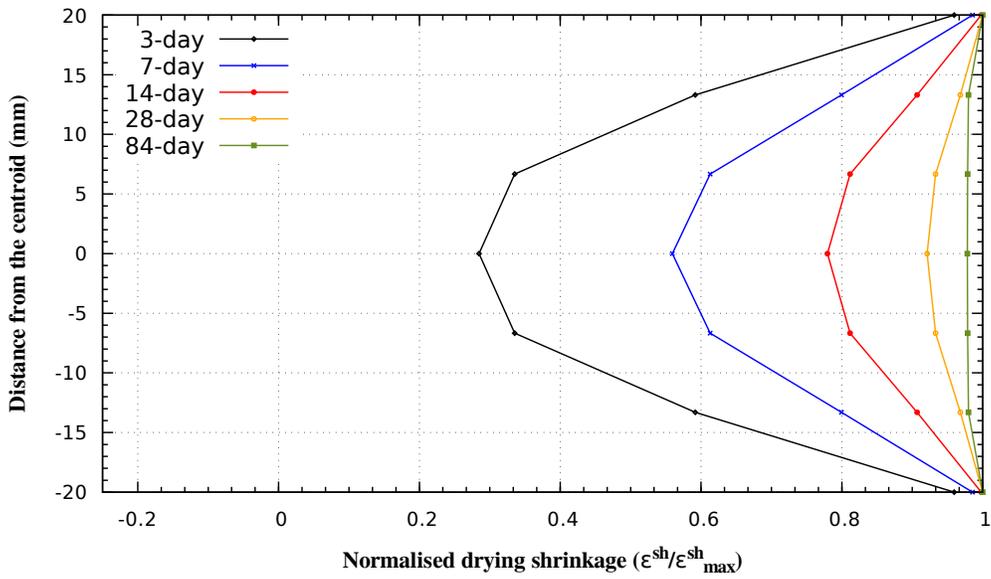
**Figure 8:** (a) Fitting of triaxial test data in meridian planes (after Ottosen [44]) and (b) uniaxial tension softening curves.



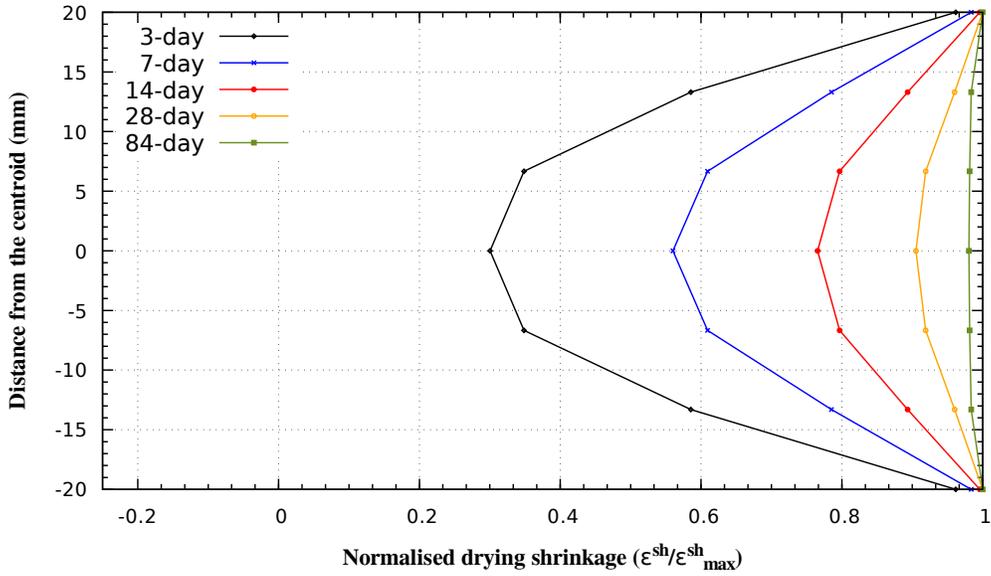
**Figure 9:** (a) Geometry and meshing of the prism model and (b) boundary conditions



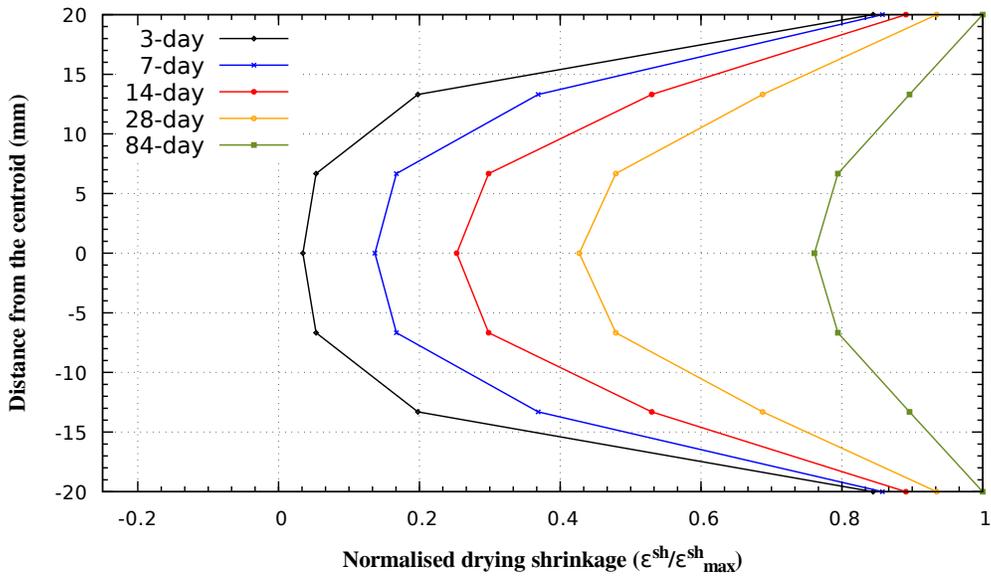
(a)



(b)



(c)

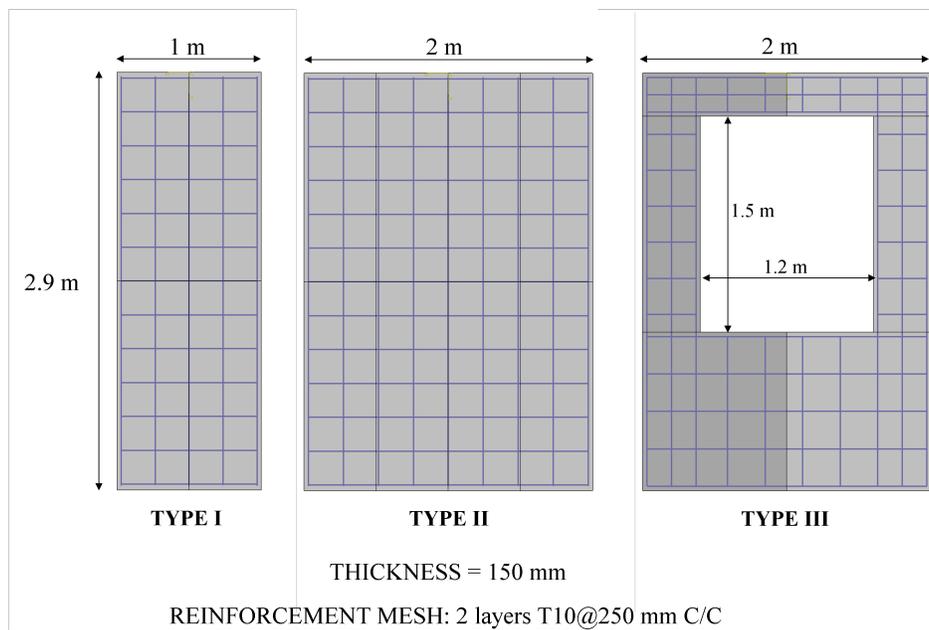


(d)

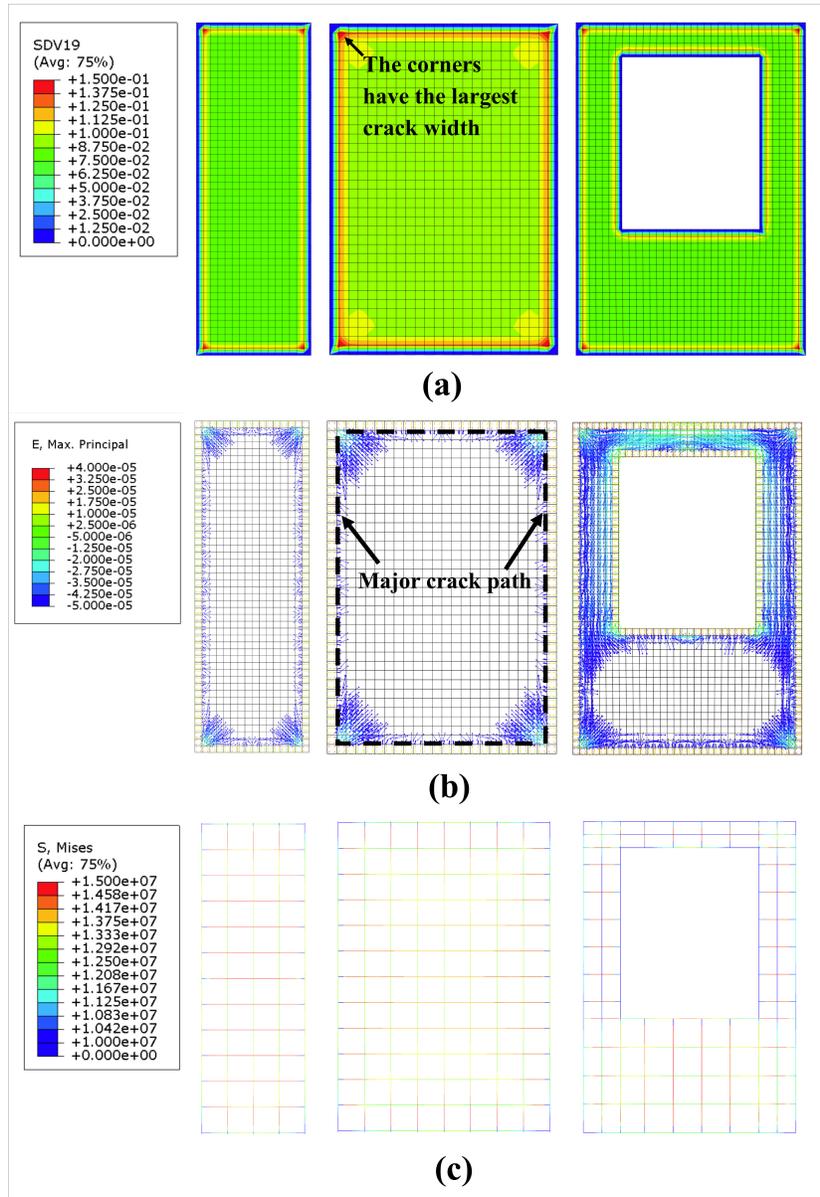
**Figure 10:** (a) Comparisons of the simulated with the tested drying-shrinkage curves, (b) cross-sectional (C-C) shrinkage distributions of GI-5, (c) cross-sectional (C-C) shrinkage distributions of GII-1, and (d) cross-sectional (C-C) shrinkage distributions of GIII-5



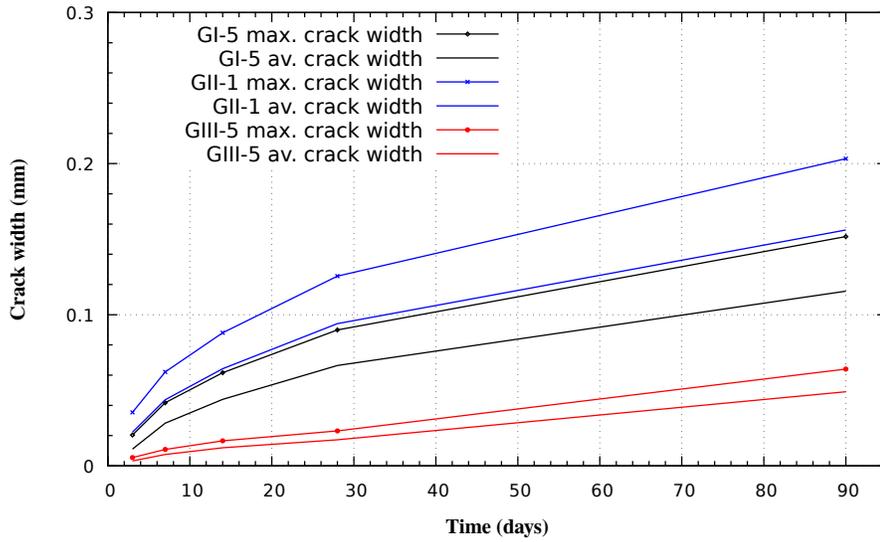
**Figure 11:** Photo of the reinforced foamed concrete slab in field test.



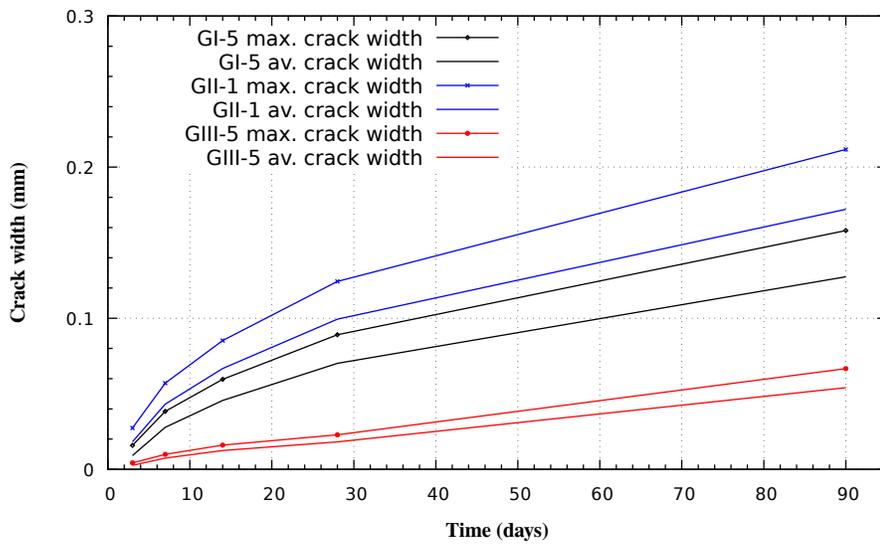
**Figure 12:** Prototypes of formworks.



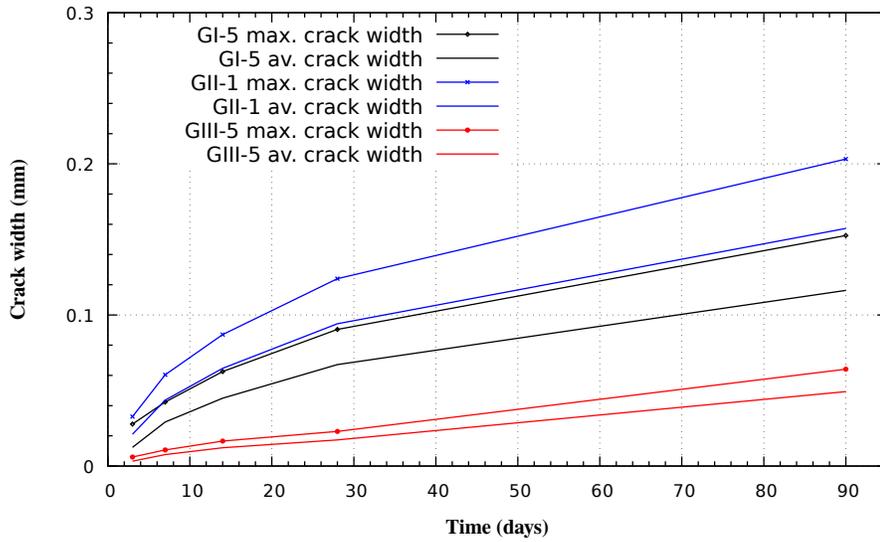
**Figure 13:** Simulated mechanical behaviour of GI-5 panel after 90 days: (a) equivalent crack width (state variable SDV 19 in mm), (b) maximum principal strain direction, and (c) von Mises distribution in steel reinforcement mesh (in Pa).



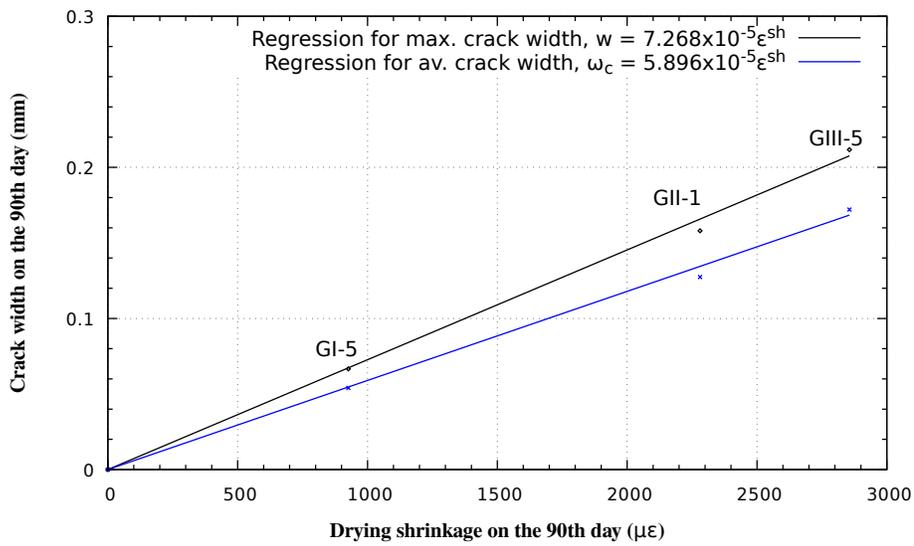
(a) Type I



(b) Type II



(c) Type III



(d)

**Figure 14:** Maximum and average crack widths: (a) Type I formwork, (b) Type II formwork, (c) Type III formwork, and (d) regression relationships between free shrinkage and crack widths developed in Type II formwork at 90 days.

**Table 1:** XRF elemental analysis of the raw materials.

	OPC	GGBS	CSA clinker	LF	SF
CaO	67.0%	46.5%	41.3%	96.9%	0.4%
SiO <sub>2</sub>	19.4%	32.2%	8.2%	1.3%	92.1%
Al <sub>2</sub> O <sub>3</sub>	3.4%	12.3%	32.9%	–	4.6%
MgO	1.0%	4.1%	2.8%	1.8%	0.5%
Fe <sub>2</sub> O <sub>3</sub>	3.5%	1.0%	1.6%	–	1.0%
SO <sub>4</sub>	5.1%	3.1%	11.8%	–	0.4%

**Table 2:** Mix proportion of non-foamed mortar and foamed concrete.

	OPC	GGBS	SF	LF	Water	FA	Wet density (kg/m <sup>3</sup> )	Compressive strength (MPa)
N1	0.667	0.333	–	–	0.285	–	2,000	94.3
F1	0.667	0.333	–	–	0.285	0.0015	1,626	43.0
N2	0.230	0.740	0.03	–	0.285	–	2,061	90.5
F2	0.230	0.740	0.03	–	0.285	0.0015	1,638	40.6
N3	0.184	0.592	0.024	0.2	0.285	–	2,060	90.0
F3	0.184	0.592	0.024	0.2	0.285	0.0015	1,611	38.0

FA = foaming agent

N = non-foamed

F = foamed

**Table 3:** Mix proportion and compressive strength of non-foamed concrete.

	OPC	GGBS	CSA-blend	SF	MEA	LF	Water	SP (%)	BA (%)	HPMC (%)	Compressive strength (MPa)
GI-1	0.97	–	–	0.03	–	–	0.285	0.125	–	0.005	105.2
GI-2	–	0.97	–	0.03	–	–	0.285	0.125	–	0.005	40.8
GI-3	0.5	0.47	–	0.03	–	–	0.285	0.125	–	0.005	96.0
GI-4	0.23	0.74	–	0.03	–	–	0.285	0.125	–	0.005	90.5
GI-5	0.77	–	–	0.03	–	0.2	0.285	0.125	–	0.005	90.1
GII-1	0.221	0.71	–	0.029	0.04 <sup>1</sup>	–	0.285	0.125	–	0.005	93.1
GII-2	0.221	0.71	–	0.029	0.04 <sup>2</sup>	–	0.285	0.125	–	0.005	99.1
GII-3	0.221	0.71	–	0.029	0.04 <sup>3</sup>	–	0.285	0.125	–	0.005	101.4
GII-4	0.221	0.71	–	0.029	0.04 <sup>4</sup>	–	0.285	0.125	–	0.005	103.3
GIII-1	–	–	0.97	0.03	–	–	0.28	0.125	–	0.005	91.7
GIII-2	–	0.74	0.23	0.03	–	–	0.30	0.125	–	0.005	35.2
GIII-3	–	–	0.77	0.03	–	0.2	0.3	0.125	0.21	0.005	72.5
GIII-4	0.115	0.54	0.115	0.03	–	0.2	0.27	0.125	0.50	0.005	59.0
GIII-5	0.115	0.425	0.23	0.03	–	0.2	0.28	0.125	0.75	0.005	76.8
GIII-6	0.115	0.225	0.23	0.03	–	0.4	0.28	0.125	0.75	0.005	48.0

1: As-received MEA

2: Calcined the as-received MEA at 800°C for 1 hour

3: Calcined the as-received MEA at 900°C for 1 hour

4: Calcined the as-received MEA at 1,000°C for 1 hour

SP = superplasticiser (solid content)

BA = Boric acid

HPMC = Hydroxypropyl methylcellulose

**Table 4:** The coefficients of best-fitted semi-log plot of drying shrinkage data and the expansive strain before shrinkage test.

Group	$a$	$c$	$R^2$	Expansive strain before shrinkage test ( $\mu\epsilon$ )
GI-1	347	1,987	0.991	23
GI-2	605	2,355	0.978	133
GI-3	466	2,439	0.984	175
GI-4	521	2,846	0.973	179
GI-5	339	1,960	0.981	103
GII-1	463	2,491	0.961	220
GII-2	522	2,936	0.959	400
GII-3	502	2,851	0.930	370
GII-4	530	3,018	0.950	316
GIII-1	176	866	0.990	276
GIII-2	154	650	0.962	260
GIII-3	246	1,040	0.990	388
GIII-4	207	905	0.948	98
GIII-5	172	614	0.898	232
GIII-6	290	1,269	0.951	164

**Table 5:** Calibrated elastoplastic parameters from drying shrinkage test.

	$f_{c,28}$ (MPa)	$E_{c,28}$ (MPa)	$\nu$	$k_t$	$G_f$ (N/m)	$\alpha_p$	$\beta_p$	$c_p$	$d_p$	$\varepsilon$	$\Psi$ (degree)
GI-5	36.0	32963.5	0.2	0.0768	139.2	1.6881	4.2545	15.115	0.9913	0.1	15
GII-1	37.2	33325.4	0.2	0.0765	140.0	1.6999	4.2769	15.185	0.9915	0.1	15
GIII-5	29.2	31254.5	0.2	0.0787	135.2	1.6303	4.1441	14.773	0.9904	0.1	15

**Table 6:** Calibrated parameters for the drying shrinkage model

	$b$ (MPa)	$K_{s,28}$	$\alpha_H$	$D_1(h = 1)$ (mm <sup>2</sup> /day)	$D_0(h = 0)$ (mm <sup>2</sup> /day)	$\alpha_D$	$\beta$	$k_H$ (mm/day)
GI-5	0.75	73962.3	1.25	12.8259	0.5772	22.22	3.8	5
GII-1	0.90	185140.9	1.36	14.5681	0.5099	28.57	3.8	5
GIII-5	0.48	33073.52	1.10	2.0463	0.0614	33.33	3.8	5

**Table 7:** Summary of the measurement of drying shrinkage in the field trial after 193 days.

Sample	Drying shrinkage ( $\mu\varepsilon$ )
Foamed GIII-5	161
90% OPC + 10% $C\bar{S}$	2,546
90% OPC + 10% $C\bar{S}H_2$	1,843
90% OPC + 5% $C\bar{S}$ + 5% $C\bar{S}H_2$	2,733